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POLYMER FILMS: GAS DIFFUSION (1973 - MAY 82)

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BIBLIOGRAPHIC INFORMATION

PB82-860719

POLYMER FILMS: GAS DIFFUSION (1973 - MAY 82)
(CITATIONS FROM THE RUBBER AND PLASTICS RESEARCH ASSOCIATION DATA BASE)

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NEW ENGLAND RESEARCH APPLICATION CENTER, STORRS, CT

NATIONAL TECHNICAL INFORMATION SERVICE, SPRINGFIELD, VA

REPORT PERIOD COVEREC: 1973 - MAY 82

THIS BIBLIOGRAPHY CONTAINS CITATIONS CONCERNING THE DIFFUSION OF GASES THROUGH POLYMERIC FILMS RELATIVE TO THE STRUCTURE AND TYPE OF GAS. EMPHASIS IS PLACED ON METHODS OF MEASURING PERMEABILITY AND THE SELECTIVE TRANSPORT OF GASES. SPECIFIC STUDIES INCLUDE PACKAGING FILMS, REINFORCED POLYMERS, RUBBERS, AND METALLIZED PLASTICS. THE PERMEABILITY OF OXYGEN AND WATER VAPOR THROUGH INSULATION ON WIRES AND CABLES IS ALSO INCLUDED. (CONTAINS 90 CITATIONS)

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SAMPLE CITATION

TITLE-----POLYPHOSPHAZENES - IMPORTANT INORGANIC POLYMERS

ACCESSION----79-01 26261L NUMBER

AUTHOR-----SINGLER, R. E. HAGNAUER, G. L. SCHNEIDER, N. S.

POLYM.NEWS,5,NO.1,SEPT.1978,P.9-17

POLYM. NEWS-5, SEPT. 1978, P.9-17

ABSTRACT----THE SYNTHESIS OF POLYPHOSPHAZENES AND THEIR CHAIN AND BULK STRUCTURES ARE DESCRIBED. THEIR USE AS FLUOROELASTOMERS AND IN FLAME-RETARDANT APPLICATIONS IS REVIEWED TOGETHER WITH THEIR BIOMEDICAL APPLICATIONS, 34 REFS. 45D.

SAMPLE SUBJECT INDEX ENTRY

ABOUT RUBBER AND PLASTICS RESEARCH ASSOCIATION

RAPRA IS THE MACHINE-READABLE DATA BASE PRODUCED BY THE RUBBER AND PLASTICS RESEARCH ASSOCIATION OF GREAT BRITAIN. THE TIME SPAN OF THE DATA BASE IS FROM JANUARY 1972 TO PRESENT. THE DATA BASE PROVIDES COVERAGE OF CHEMISTRY AND CHEMICAL ENGINEERING OF POLYMERS (RUBBERS AND PLASTICS). TECHNICAL AND COMMERCIAL ASPECTS OF RUBBER AND PLASTICS ARE COVERED.

THE SOURCES OF THE CITATIONS ARE APPROXIMATELY 60% JOURNAL ARTICLES, 25% PATENTS AND THE REMAINDER FROM MONOGRAPHS, TRADE LITERATURE AND GOVERNMENT REPORTS. ABOUT 20,000 ITEMS ARE ADDED TO THE DATA BASE ANNUALLY, AND THE DATA BASE PRESENTLY CONTAINS ABOUT 150,000 CITATIONS. ABOUT 10 JOURNALS ARE ABSTRACTED IN THEIR ENTIRETY, WITH SELECTIVE ABSTRACTING PERFORMED FROM 400 JOURNAL TITLES REVIEWED FOR INPUT.

ABOUT PUBLISHED SEARCHES

PUBLISHED SEARCHES ARE SPECIAL INFORMATION PRODUCTS DEVELOPED FROM A VARIETY OF ONLINE DATA BASES. THE NTIS DATA BASE, WHICH IS THE KEYSTONE OF THE PUBLISHED SEARCH PROGRAM, ALONE CONTAINS MORE THAN 750,000 DOCUMENT/DATA RECORDS OF GOVERNMENT-SPONSORED RESEARCH. OTHER DATA BASES SEARCHED INCLUDE THOSE OF THE U.S. FIRE ADMINISTRATION, FEDERAL EMERGENCY MANAGEMENT AGENCY; AMERICAN PETROLEUM INSTITUTE; U.S. DEPARTMENT OF ENERGY(EDB); ENGINEERING INDEX; INSTITUTE OF PAPER CHEMISTRY; MANAGEMENT CONTENTS; INFORMATION RETRIEVAL LIMITED; INSTITUTION OF ELECTRICAL ENGINEERS (INSPEC); INTERNATIONAL FOOD INFORMATION SERVICE (FSTA); AND THE NATIONAL AERONAUTICS AND SPACE ADMINISTRATION. (IAA).

PUBLISHED SEARCHES ARE SPECIALLY PREPARED BIBLIOGRAPHIES REFERENCING REPORTS WITH FULL BIBLIOGRAPHIC CITATIONS, INCLUDING INFORMATIVE ABSTRACTS AND, WHEN POSSIBLE, ORDERING INFORMATION AND PRICE. THE ABSTRACTS PROVIDE A QUICK, INEXPENSIVE WAY TO DETERMINE WHICH REPORTS IN THE NTIS DATA BASE, FOR ONE, ARE OF SPECIAL INTEREST TO A USER. THE SEARCHES ARE PREPARED BY INFORMATION SPECIALISTS AND ARE AVAILABLE IN MANY TOPIC AREAS; THEY ARE UPDATED AT REGULAR INTERVALS, AND COST THIRTY DOLLARS IN PAPER OR MICROFICHE FOR DOMESTIC ORDERS. A COMPLETE LIST OF CURRENT PUBLISHED SEARCHES IS AVAILABLE BY REQUESTING BROCHURE NUMBER PB82-105024 FOR FIVE DOLLARS, REFUNDABLE WITH FIRST PUBLISHED SEARCH PURCHASE. IN ADDITION TO REGULAR UPDATING, NEW TITLES (SEARCHES) ARE BEING ADDED EACH WEEK AND NEW ARRANGEMENTS ARE BEING COMPLETED WITH DATA BASE OWNERS TO ACCESS AN EVEN GREATER VARIETY OF INFORMATION SOURCES.

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CITATIONS

SORPTION AND DIFFUSION OF HYDROCARBON VAPOURS IN GLASSY POLYMERS 82-03 03929L

BARRIE, J. A. WILLIAMS, M. J. L. MUNDAY, K.

POLYM.ENGNG.SCI.,20,NO.1,MID-JAN.1980,P.20-9

SORPTION ISOTHERMS IN THE REGION OF LOW RELATIVE PRESSURES WERE DETERMINED AT SEVERAL TEMPERATURES FOR METHANE, PROPANE AND CHLORODIFLUORCMETHANE IN PS, AND FOR PROPANE IN BISPHENOL A POLYCARBONATE AND POLYVINYL ACETATE. THE RESULTS WERE WELL REPRESENTED BY THE ISOTHERM EQUATION OF DUAL SORPTION THEORY AS APPLIED TO GLASSY POLYMERS. A STUDY WAS MADE OF THE TEMPERATURE DEPENDENCE OF THE ISOTHERM PARAMETERS. THE LANGMUIR COMPONENT TO SORPTION DECREASED AS TO WAS APPROACHED, AND MEASUREMENTS WITH POLYVINYL ACETATE CONFIRMED THAT THIS COMPONENT WAS ABSENT ABOVE THE TRANSITION. AVERAGE DIFFUSION COEFFICIENTS WERE OBTAINED FROM SORPTION (DESORPTION) RATE CURVES AT CONSTANT PRESSURE FOR PROPANE IN PS AND POLYCARBONATE, AND A PROCEDURE WAS DEVELOPED FOR THEIR ANALYSIS TO YIELD THE DIFFUSION COEFFICIENTS OF THE SORBED SPECIES. 24 REFS. 93513

PRESSURE DEPENDENCE OF DIFFUSION COEFFICIENT FOR CARBON DIOXIDE IN GLASSY POLYMERS 82-03 03930L

TOI, K.

POLYM.ENGNG.SCI.,20,NO.1,MID-JAN.1980,P.30-5

THE PRESSURE DEPENDENCE OF THE APPARENT DIFFUSION AND PERMEATION COEFFICIENTS WAS OBSERVED USING THE PERMEATION TIME-LAG METHOD FOR CARBON DIOXIDE IN GLASSY PETP, PS AND PVC BELOW 1 ATM. THE PERMEATION COEFFICIENT WAS CONSTANT, WHEREAS THE DIFFUSION COEFFICIENT INCREASED WITH PRESSURE. IT WAS CONCLUDED THAT THE ADSORBED CARBON DIOXIDE WAS COMPLETELY IMMOBILISED AND DID NOT PARTICIPATE DIRECTLY IN THE DIFFUSION. A COMPUTER WAS USED IN THE NUMERICAL CALCULATION TO DETERMINE THE TRUE DIFFUSION COEFFICIENT FROM THE MODEL OF PAUL ET AL. 10 REFS. 93513

FICKIAN DIFFUSION OF ALKANES THROUGH GLASSY POLYMERS: EFFECTS OF TEMPERATURE, DIFFUSANT SIZE AND POLYMER STRUCTURE 82-03 03931L

CHEN, S. P. EDIN, J. A. D.

POLYM.ENGNG.SCI., 20, NO.1, MID-JAN. 1980, P. 40-50

DIFFUSIVITIES (D) RANGING OVER SIX ORDERS OF MAGNITUDE WERE OBTAINED USING A PERMEATION APPARATUS EMPLOYING A GAS FLOW METHOD AND A FLAME IONISATION DETECTOR. LOG D FOR HYDROCARBONS IN BISPHENDL A POLYCARBONATE (PC) AT 120C WAS PROPORTIONAL TO THE SQUARE OF THE MOLECULAR DIAMETER AS GIVEN BY THE LENNARD-JONES POTENTIAL. THIS CORRELATION HELD EVEN FOR THE NON-SPHERICAL N-HEXANE MOLECULE. ACTIVATION ENERGY FOR DIFFUSION WAS ALSO LINEARLY RELATED TO MOLECULAR DIAMETER, WITH VALUES OF 9.5 AND 23 KCAL/MOL FOR METHANE AND NEOPENTANE IN PC, RESPECTIVELY. THE DIFFUSION DATA DID NOT CORRELATE WITH THE TG OF SIMILAR POLYMERS OF HIGHER TG, BUT THE PRESENCE OF SUBSIDIARY TRANSITIONS APPEARED TO ENHANCE SEGMENTAL MOBILITY, INCREASING THE HYDROCARBON RATE OF DIFFUSION. 32 REFS. 43C12-93513

STATISTICAL MECHANICAL MODEL OF SORPTION AND DIFFUSION OF SIMPLE PENETRANTS IN POLYMERS 82-03 03932L

PACE, R. J. DATYNER, A.

POLYM.ENGNG.SCI.,20,NO.1,MID-JAN.1980,P.51-8

TWO POSSIBLE TYPES OF RANDOM MOTION FOR A SPHERICAL PENETRANT IN AN AMORPHOUS POLYMER ARE DESCRIBED, ONE TYPE DETERMINING THE JUMP FREQUENCY AND ACTIVATION ENERGY OF DIFFUSION, THE OTHER DETERMINING THE JUMP LENGTH. SORPTION OF SIMPLE GASES AT LOW PENETRANT PRESSURES IS ASSUMED TO OCCUR MOSTLY IN PRE-EXISTING HOLES, BOTH ABOVE AND BELOW TG, AND THE SAME PENETRANT DIFFUSION MECHANISM IS ASSUMED TO HOLD IN THE TWO REGIONS. CHANGES IN APPARENT HEAT OF SOLUTION AND ACTIVATION ENERGY OF DIFFUSION OBSERVED AT TG ARE EXPLAINED IN TERMS OF ADDITIONAL HOLE FORMATION WITH INCREASE IN TEMPERATURE ABOVE TG. EVIDENCE IS GIVEN THAT HOLE FORMATION IN SIMPLE POLYMERS SUCH AS PE MAY OCCUR BY CHAIN KINKING, WHILE FOR POLYMERS WITH ARTICULATED SIDE GROUPS IT APPEARS THAT HOLE FORMATION ARISES PRINCIPALLY FROM MOTIONS WITHIN THESE GROUPS. 26 REFS. 93513

SELECTION OF BARRIER MATERIALS FROM MOLECULAR STRUCTURE 82-03 03934L

LEE, W. M.

POLYM.ENGNG.SCI.,20,NO.1,MID-JAN.1980,P.65-9

A PREDICTION TECHNIQUE FOR GAS PERMEABILITY FROM POLYMER MOLECULAR STRUCTURE WAS DEVELOPED ON THE BASIS OF A SPECIFIC FREE VOLUME DIFFUSION THEORY, IN WHICH THE FREE VOLUME AVAILABLE PER UNIT MASS IN A POLYMER STRUCTURE CONTROLS THE RATE OF DIFFUSION AND, HENCE, THE RATE OF PERMEATION OF THE GAS. THE THEORY PREDICTS A LINEAR RELATIONSHIP BETWEEN LOG PERMEABILITY AND SPECIFIC VOLUME, AND A NUMBER OF POLYMERS COVERING SIX ORDERS OF MAGNITUDE IN CARBON DIOXIDE AND DXYGEN PERMEABILITY FOLLOWED THIS CORRELATION. THIS TECHNIQUE GREATLY SIMPLIFIES THE SELECTION OF BARRIER MATERIALS FOR PACKAGING APPLICATIONS. MOLECULAR STRUCTURES WITH STRONG POLAR-TO-POLAR INTERACTIONS AND HYDROGEN BONDING FORCES PROVIDE GCCD BARRIERS TO CARBON DIOXIDE AND OXYGEN. 13 REFS. 93513

DXYGEN TRANSMISSION THROUGH HIGHLY CROSSLINKED POLYMERS 82-03 03935L

GORDON, G. A. RAVVE, A.

POLYM.ENGNG.SCI.,20,NO.1,MID-JAN.1980,P.70-7

HIGHLY CROSSLINKED POLYMERS OF VARYING STRUCTURE WERE PRODUCED BY REACTION OF POLYGLYCIDYL ACRYLATE AND POLYGLYCIDYL METHACRYLATE WITH CHLORENDIC, GLUTARIC, MALEIC, SUCCINIC AND POLYMALONIC ANHYDRIDE CROSSLINKING AGENTS. TOPOLOGY WAS VARIED BY THE USE OF A DILUENT AND A COMONOMER IN THE BACKBONE CHAIN. DXYGEN PENETRATION MEASUREMENTS WERE MADE ON THESE POLYMERS COATED ONTO A PP FILM BEFORE CROSSLINKING. THE CROSSLINKING PROCESS GREATLY REDUCED THE DXYGEN PERMEABILITY WHICH, HOWEVER, WAS DEPENDENT NOT ONLY ON THE DEGREE OF CROSSLINKING BUT ALSO ON THE CROSSLINK DENSITY, THE CHEMICAL NATURE OF THE STRUCTURAL ELEMENTS, AND THE TOPOLOGY OF THE POLYMER NETWORK. 27 REFS. 42C35-55CAH-93513

VAPOUR-PHASE POLYMERISATION : I. FORMATION OF POLY (P-PHENYLENETEREPHTHALAMIDE GAS BARRIER COATINGS 82-02 02023L

IKEDA, R. M. ANGELO, R. J. BOETTCHER, F. P. BLOMBERG, R. N. SAMUELS, M. R.

DRG.CDATINGS PLAST.CHEM., VOL.41, SEPT.1979, P.287-92

THIN COATINGS OF POLYPHENYLENE TEREPHTHALAMIDE WERE SYNTHESISED BY VAFOUR PHASE POLYCONDENSATION AND CONCURRENT DEPOSITION ONTO A PET SUBSTRATE. THE COATINGS HAD GOOD RESISTANCE TO PERMEATION BY OXYGEN AND WERE ALSO MOISTURE RESISTANT. 5 REFS. 43C318-6A31-723

INVERSE GAS CHROMATOGRAPHY, A METHOC FOR STUDYING POLYMER-MIGRANT INTERACTIONS IN POLYOLEFIN PACKAGING MATERIALS 82-C2 02270L

SENICH, G. A. SANCHEZ, I. C.

DRG.CDATINGS PLAST.CHEM., VOL.41, SEPT.1979, P.345-9

THE USE OF INVERSE PHASE GAS CHROMATOGRAPHY TO MEASURE THE EXTENT OF MIGRATION OF AN IMPURITY OR ADDITIVE FROM A POLYMERIC CONTAINER INTO ITS CONTENTS IS DISCUSSED. THE SPECIFIC RETENTION VOLUME CAN BE USED TO CALCULATE THE POLYMER-MIGRANT INTERACTION PARAMETERS WHICH WHEN COMBINED WITH SOLUBILITY OF THE MIGRANT IN THE SOLVENT, ALLOWS THE EQUILIBRIUM PARTITION COEFFICIENT TO BE CALCULATED. THE DIFFUSION COEFFICIENT OF MIGRANT IN POLYMER CAN BE USED TO ESTIMATE THE DEGREE OF MIGRATION DURING THE EXPECTED SHELF LIFE. THERMODYNAMIC AND KINETIC EFFECTS ARE TAKEN INTO CONSIDERATION. 19 REFS. 42C11-935T

EFFECT OF FILLERS ON THE PROPERTIES OF URETHANE POLYMERS 82-01 00264L

MAIK, M. KRYSZTAFKIEWICZ, A.

POLIM.TWORZ.WIELK.,26,NO.7,JULY 1981,P.245-7 LANG-POLISH

THE DEPENDENCE OF POLYMER PROPERTIES ON FILLER CONTENT, ON COMPOSITION, THICKNESS AND DENSITY OF THE INTERPHASE LAYER AND ON SURFACE CHARACTERISTICS OF FILLER WAS INVESTIGATED. PARTICULAR ATTENTION WAS PAID TO SILICA FILLERS AND THEIR INFLUENCE ON THE FROPERTIES, E.G. MCRPHOLOGY AND GAS PERMEABILITY, OF URETHANE POLYMERS. 35 REFS. 43C6-51SS-9 ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

MODEL FOR PERMEATION OF MIXED GASES AND VAPOURS IN GLASSY POLYMERS 82-01 00520L

KOROS, W. J. CHERN, R. T. STANNETT, V. HOPFENBERG, H.

J.POLYM.SCI.POLYM.PHYS., 19, NO.10, OCT. 1981, P. 1513-30

A MODEL WAS DISCUSSED WHICH EXPLAINED COMPLEX EFFECTS OF FEED COMPOSITION AND PRESSURE ON COMPONENT PERMEABILITIES IN HIGH-PRESSURE GAS SEPARATORS BASED ON GLASSY POLYMER MEMBRANES. PERMEATION OF GAS MIXTURES WAS ANALYSED ON THE BASIS OF A FORM OF FICK'S LAW WHICH ACCOUNTS FOR THE FACT THAT PENETRANTS IN GLASSY POLYMERS SORB INTO AND DIFFUSE THROUGH TWO DIFFERENT MOLECULAR ENVIRONMENTS. POTENTIAL DEVIATIONS FROM THE THEORY WERE DISCUSSED IN TERMS OF SEPARABLE EFFECTS DUE TO GAS SOLUBILITY AND MOLECULAR MOBILITY. 47 REFS. 6M-93513

INFLUENCE OF PETRCLEUM GAS ON THE DIFFUSION PROPERTIES OF POLYMERIC MEMBRANES 81-12 79071L

OLENINA, Z. K. ERSHOV, B. N. BRESHCHENKO, E. M.

PLAST.MASSY, NO.5, 1980, P.29 LANG- RUSSIAN

A STUDY IS DESCRIBED OF THE PERMEABILITY OF A MEMBRANE MADE FROM POLYPROPYLENE AND A FLUORINE-CCNTAINING PLASTIC UNIRRADIATED AND IRRADIATED WITH ARGON IONS AFTER PROLONGED TREATMENT WITH PETROLEUM GAS. 5 REFS. 42C12-6M-93513 ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

GAS PERMEATION THROUGH POROUS POLYMERIC MEMBRANE 81-11 77873L

KAMIDE, K. MANABE, S. NOHMI, T. MAKINO, H. KAWAI, T.

POLYM.PREPRINTS, 21,NO.2, AUG. 1980, P.90-2

THE GAS FLOW MECHANISM IN POLYMERIC MEMBRANES HAVING VARIOUS TYPES OF PORE SIZE DISTRIBUTION WAS STUDIED. 11 REFS. 6M-93513

SILICONE RUBBER TECHNOLOGY REVIEW 81-11 76608C

POLMANTEER, K. E.

MINNEAPOLIS, MINN., JUNE 2-5, 1981, PAPER 1, PP.64. PREPRINT. 012 PUBLCN. DETAILS - ACS, RUBBER DIV. 119TH MEETING, SPRING 1981. PAPERS

A REVIEW IS PRESENTED OF THE CHEMICAL STRUCTURE, SYNTHESIS, VULCANISATION, SOLUBILITY, SURFACE ENERGY, PERMEABILITY, OXIDATION, HYDROLYSIS AND RHEOLOGICAL AND LOW TEMPERATURE PROPERTIES OF SILICONE RUBBERS, INCLUDING POLYDIMETHYLSILOXANE AND SILOXANE COPOLYMERS. FILLERS AND THEIR EFFECTS ON MECHANICAL PROPERTIES ARE ALSO CONSIDERED. 28 REFS. 45C

PERMEATION OF GASES THROUGH MODIFIED POLYMER FILMS. V. PERMEATION AND DIFFUSION OF HELIUM, NITROGEN, METHANE, ETHANE AND PROPANE THROUGH GAMMA-RAY CROSSLINKED POLYETHYLENE 81-11 76102L

MACDONALD, R. W. HUANG, R. Y. M.

J.APPL.POLYM.SCI.,26,NO.7,JULY 1981,P.2239-63

GAMMA-IRRADIATION OF PE FILMS WAS CARRIED OUT IN VACUUM, ACETYLENE AND ACETYLENE-NITROGEN MIXTURES IN ORDER TO STUDY THE CHANGES INDUCED IN EFFICIENCY OF CROSSLINKING AND TRANSPORT PROPERTIES WITH IRRADIATION DOSE AND ATMOSPHERE. PERMEABILITY AND DIFFUSION CONSTANTS WERE OBTAINED FOR HELIUM, NITROGEN, METHANE, ETHANE AND PROPANE AND THE PLASTICISING EFFECT OF THE HYDROCARBON GASES INVESTIGATED. 50 REFS. 42C11-8953-93513

POLYMER BREAKS THE GAS BARRIER 81-10 75169L

PLAST.RUBB.WKLY.,NO.901,22ND AUG.1981,P.5 CORP. AUTH-KANEGAFUCHI CHEMICAL INDUSTRY CO.

BRIEF DETAILS ARE PRESENTED ON KANEKA PARNEX, AN AMORPHOUS ACRYLIC COPOLYMER, WHICH IS REPORTED TO HAVE EXCELLENT CHEMICAL RESISTANCE AND HIGH GAS BARRIER PROPERTIES. THE MATERIAL CAN BE INJECTION, EXTRUSION OR BLOW MOULDED, VACUUM FORMED OR CALENDERED. APPLICATIONS, INCLUDING PACKAGING, PUMPS, ELECTRICAL COMPONENTS AND PETROL CAPS, ARE MENTIONED. 42C351A-9351

DIFFUSION OF GASES IN GLASSY POLYMERS. II. DUAL SORPTION THEORY 81-07 69357L

NAPP, S. J. PEPPAS, N. A.

POLYM.NEWS,7,NO.4,1981,P.174-7

THE ABILITY OF THE DUAL MODE THEORY TO PREDICT AND CORRELATE A RANGE OF MASS TRANSPORT PROCESSES WHERE ONE POPULATION IS IMMOBILE RELATIVE TO ANOTHER IS EXAMINED. 17 REFS. 93513

COMPARATIVE DERIVATOGRAPH TESTING OF THE THERMAL DECOMPOSITION OF SUSPENSION PVC AND PVC POWDER POLYMERISED IN BULK 81-06 68247L

BODRI, E. KOVACS, B. ZIEGLER, B.

MUANYAG ES GUMI., 18, NO.3, 1981, P.85-7 LANG- HUNGARIAN

DERIVATOGRAPH RECORDS SHOW THAT THE THERMAL DECOMPOSITION OF PVC PRODUCED BY BULK POLYMERISATION AND SUSPENSION POLYMERISATION IS EXOTHERMIC AND ENCOTHERMIC, RESPECTIVELY. THE DIFFERENCE IS DUE TO THE PERICELLULAR SURFACE SKIN FOUND ON SUSPENSION PVC GRANULES WHICH HINDERS THE ACCESS OF OXYGEN, WHILE IN THE CASE OF BULK PVC GRANULES OXYGEN IS ABLE TO CONTACT PRIMARY GRANULES UNRESISTED. 9 REFS. 42C382-622-932 ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

EFFECT OF TENSILE DEFORMATIONS ON CASE TRANSPORT IN GLASSY POLYMER FILMS 81-06 67207L

SMITH. T. L. ADAM, R. E.

POLYMER, 22, NO.3, MARCH 1981, P.299-304

THE EFFECT OF SIMPLE TENSILE DEFORMATIONS ON THE PERMEABILITY AND DIFFUSION COEFFICIENTS OF GASES IN GLASSY POLYMERS WAS STUDIED. THE FILMS USED WERE POLYCARBONATE AND POLYIMIDE WITH CARBON DIOXIDE AND NITROGEN. FOR NITROGEN IN PC THE COEFFICIENTS DECREASED WITH TIME AT CONSTANT STRAIN AND WERE PARTIALLY DEPENDENT ON STRAIN AND THERMAL HISTORIES. FOR CARBON DIOXIDE IN POLYIMIDE, THE COEFFICIENTS INCREASED WITH STRAIN UP TO 2% AT 125C AND THEN DECREASED, SHOWING A STRAIN-INDUCED RELAXATION. 41 REFS. 43C12-93513

EFFECT OF VARIOUS FACTORS ON THE PERMEABILITY OF GASES THROUGH POLYMER FILMS. III. PROPERTIES OF FILMS PREPARED FROM MODIFIED POLYSTYRENE AND PVC 81-05 65408L

KAMINSKA, A.

POLIM.TWORZ.WIELK.,25,NO.12,1980,P.442-4 LANG- POLISH

THE EFFECT OF METHYL METHACRYLATE-BUTADIENE-STYRENE COPOLYMERS (MBS) AND SAN AS MODIFIERS ON THE PERMEABILITY TO AIR OF FILMS OF PVC AND PS, AS WELL AS ON THE UV STABILITY OF SUCH FILMS, WAS INVESTIGATED. ADDITION OF 1 TO 3% MBS OR SAN WAS FOUND TO REDUCE THE PERMEABILITY OF PS AND INCREASE THAT OF PVC; RESISTANCE TO UV IRRADIATION WAS REDUCED IN BOTH POLYMERS. PERMEABILITY MEASUREMENT APPEARED TO BE A SENSITIVE METHOD FOR DETECTING CHANGES IN THE STRUCTURE OF POLYMERS. 5 REFS. 93513 ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

DIFFUSION OF GASES IN GLASSY POLYMERS. I. FREE VOLUME AND ENERGY FLUCTUATION THEORIES 81-05 66434L

NAPP, S. J. PEPFAS, N. A.

POLYM.NEWS,7,NO.3,JAN.1981,P.118-21

THE FREE VOLUME AND ENERGY FLUCTUATION THEORIES WHICH DESCRIBE THE KINETICS OF GASEOUS TRANSPORT IN HOMOPOLYMERS ARE BRIEFLY DISCUSSED. A MORE DETAILED DISCUSSION OF THE DUAL SORPTION THEORY IS ALSO GIVEN. 12 REFS. 93513

BUTYL RUBBER AND ITS USE IN NON-TYRE APPLICATIONS 81-02 59593L

RITCHIE, K.

HULE MEX.PLAST., 36,NO.414, JULY 1980, P.5/18 LANG-SPANISH

THE CHEMICAL STRUCTURE AND PREPARATION OF BUTYL RUBBER AND

HALOBUTYL RUBBERS IS DISCUSSED, AND THE MAIN PROPERTIES OF THESE MATERIALS ARE REVIEWED, I.E. GAS PERMEABILITY, THERMAL STABILITY, OZONE RESISTANCE, ENERGY ABSORPTION AND CHEMICAL RESISTANCE. ATTENTION IS ALSO PAID TO COMPOUNDING, CURING SYSTEMS AND CTHER COMPOUNDING INGREDIENTS. APPLICATIONS EXAMINED INCLUDE PHARMACEUTICAL STOPPERS, VEHICLE SUSPENSIONS, TANK LININGS, MEMBRANES, ELECTRICAL CONDUCTORS, POLYMERIC PROPERTY MODIFIERS FOR PLASTICS, FOOTBALL BLADDERS, SEALANTS AND CONVEYOR BELTING. FORMULATIONS FOR THESE APPLICATIONS ARE GIVEN, WITH PARTICULAR REFERENCE TO POLYMERS PRODUCED BY POLYSAR LTD. 7 REFS. 42C131D12

SELECTIVE TRANSPORT OF SULPHUR DIOXIDE THROUGH POLYMER MEMBRANES. I. POLYACRYLATE AND CELLULOSE TRIACETATE SINGLE LAYER MEMBRANES 81-01 58208L

KUEHNE, D. L. FRIEDLANDER, S. K.

TEC PROCESS DES.CEV., 19, NO.4, DCT. 1980, P. 609-16

COMMERCIAL SAMPLES OF POLYACRYLATE AND CELLULOSE TRIACETATE POLYMERS WERE FOUND TO BE SELECTIVELY PERMEABLE TO SULPHUR DIOXIDE GAS. ULTRA THIN FILMS WERE PREPARED TO ACHIEVE HIGH SO2 FLUXES, REQUIRED FOR SO2 GAS SEPARATORS. SO2 PERMEABILITY HAD A LARGE PRESSURE DEPENDENCE, WHICH ADVERSELY AFFECTED MEMBRANE PERFORMANCE. 18 REFS. 6M-93513

SELECTIVE TRANSPORT OF SULPHUR DIOXIDE THROUGH POLYMER MEMBRANES. II. CELLULOSE TRIACETATE/POLYACRYLATE COMPOSITE MEMBRANES 81-C1 58209L

KUEHNE, D. L. FRIEDLANDER, S. K.

IEC PROCESS DES.DEV., 19, NO.4, DCT. 1980, P. 616-23

A NOVEL TECHNIQUE IS DESCRIBED FOR PREPARING POLYACRYLATE/CELLULOSE TRIACETATE POLYMER MEMBRANES FOR SOZ GAS SEPARATIONS. SOZ PERMEATION DATA WERE OBSERVED TO BE PRESSURE DEPENDENT. A CONTINUOUS FLOW SYSTEM WAS BUILT TO MODEL THE GAS FLOW IN A MEMBRANE SEPARATOR, AND SELECTED MEMBRANES WERE TESTED WITH BINARY MIXTURES OF NZ AND SOZ. MEMBRANE PERFORMANCE WAS EVALUATED IN TERMS OF DESIGN GOALS FOR COMBUSTION AND SMELTER GAS APPLICATIONS. 6 REFS.

EFFECT OF VARIOUS FACTORS ON THE PERMEABILITY OF GASES THROUGH POLYMER FILMS. II. CHANGES IN THE PERMEABILITY OF AIR THROUGH POLYSTYRENE FILMS. UNDER THE EFFECT OF UV IRRADIATION 80-12 57154L

KAMINSKA, A. MLYNARCZYK, D.

POLIM.TWORZ.WIELK,25,NO.8,AUG.1980,P.289-90

PHOTOCHEMICAL REACTIONS TAKING PLACE IN PS, WHICH POLISH. CHANGE THE MOLECULAR STRUCTURE AND THE STRUCTURE OF THE FILMS, AFFECT THE PERMEABILITY OF PS FILMS. CROSSLINKING DECREASES, AND DEGRADATION AND BRANCHING INCREASE THE THE DIFFERENCES IN CHANGES IN PERMEABILITY OF PERMEABILITY. PS FILMS OF DIFFERENT MOLECULAR WEIGHT AND POLYMOLECULARITY SUGGEST THAT THE COURSE OF PHOTOCHEMICAL PROCESSES DEPENDS ON BOTH THESE PARAMETERS. A DECREASE IN RESISTANCE TO THE ACTION OF UV IRRADIATION IS OBSERVED WITH INCREASING PS POLYMOLECULARITY. IT IS CONCLUDED THAT THE DETERMINATION OF CHANGES IN PERMEABILITY OF FILMS MAY BE USED TO EXAMINE THE COURSE OF PHOTOCHEMICAL PROCESSES IN POLYMER FILMS. 5 REFS. ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

GAS PERMEABILITY OF HIGH POLYMERS PARTICULARLY OF RUBBERS 80-12 57566L

SCHUCK. H.

KAUT.U.GUMMI KUNST.,33,NO.9,SEPT.1980,P.705-15

GERMAN. THE RUBBERS STUDIED INCLUDE POLYISOBUTYLENE, SILICONE RUBBER AND NR. 17 REFS.

POLYMERISATION FROM THE VAPOUR PHASE. I.
POLY-P-PHENYLENETEREPHTHALAMIDE (PPTA) GAS BARRIER
CDATINGS 80-11 55870L

IKEDA, R. M. ANGELO, R. J. BOETTCHER, F. P. BLOMBERG, R. N. SAMUELS, P. R.

J.APPL.POLYM.SCI., 25, NO.7, JULY 1980, P.1391-405

AN ATMOSPHERIC-PRESSURE, VAPOUR-PHASE POLYMERISATION TECHNIQUE WAS USED TO DEPOSIT THIN PPTA COATINGS ONTO SHEET SUBSTRATES. A MINIMUM DEPOSITION TEMP. OF 170C WAS FOUND TO BE CRITICAL. THESE COATINGS EXHIBITED GOOD DXYGEN BARRIER PROPERTIES AND WERE FOUND TO CONSIST OF FUSED 0.1 MU M PARTICULATES. SEM OF THE TOP SURFACES OF THE COATINGS REVEALED THEIR PARTICULATE ORIGIN. SIMILAR EVIDENCE WAS ALSO OBTAINED FROM SEM EXAMINATION OF FRACTURE SURFACES AND TRANSMISSION ELECTRON MICROSCOPY OF MICROTOMED SECTIONS. THE COALESCENCE OF THESE COATINGS WAS SHOWN BY SEM OF PLASMA-ETCHED SURFACES AND DXYGEN PERMEABILITY DATA. THE UNIPLANAR ORIENTATION OF THE POLYMER CRYSTALS IN THESE COATINGS WAS STRONG EVIDENCE FOR THE EPITAXIAL GROWTH OF THE CRYSTALS. 5 REFS.

POLYMER MEMBRANE ELECTRODE BASED POTENTIOMETRIC AMMONIA GAS SENSOR 8C-11 55922L

MEYERHOFF, M. E.

ANALYT. CHEM.,52, NC.9, AUG.1980, P.1532-4

A DESCRIPTION IS GIVEN OF THE INCORPORATION OF THE ANTIBIOTIC NONACTIN INTO AN APPROPRIATE PLASTICISER/PVC MEMBRANE ELECTRODE AND THEN USED AS THE INNER ELECTRODE IN A GAS SENSING ARRANGEMENT FOR DISSOLVED AMMONIA. 7 REFS.

POSSIBLE MECHANISM OF SELECTIVITY OF DIFFUSION OF MOLECULAR GASES IN POLYMERS 80-11 54673L

BRUEV, A. S.

POLYM.SCI.USSR, 21, NO. 5, 1979, P. 1256-66

A STUDY WAS MADE OF THE DIFFUSION OF NON-SPHERICAL MOLECULES IN POLYMER FILMS. IT WAS SHOWN THAT THE CORRELATION BETWEEN THE DIRECTION OF CHANGE IN DIFFUSION AND MOLECULAR ORIENTATION IN THE CASE OF DIATOMIC AND LINEAR MOLECULES COULD INCREASE THE RATE OF DIFFUSION TRANSFER. RESULTS WERE IN AGREEMENT WITH EXPERIMENTAL DATA CONCERNING THE DIFFUSION OF MOLECULAR GASES IN POLYMERS. 25 REFS.

ELASTOMERE AUF BASIS ATHYLEN-PROPYLEN 80-10 53121A

42C11C12 CORP. AUTH- WIRTSCHAFTSVERBAND DER DEUTSCHER KAUTSCHUKIND. PUBLCN. DETAILS- GRUNES BUCH 39

GERMAN. THIS BOOK CONTAINS SECTIONS ON THE POLYMERISATION OF ETHYLENE-PROPYLENE RUBBER, MONOMERS, CATALYSTS AND POLYMERISATION KINETICS; THE MOLECULAR STRUCTURE AND MORPHOLOGY OF THE ELASTOMERS, E.G. MWD, TG, CRYSTALLINITY, RHEOLOGICAL PROPERTIES, THE VULCANISATION OF EPM AND EPDM, CURING AGENTS AND CROSSLINKING REACTIONS, PROCESSING, EXTRUSION, CALENDERING, MIXING, FILLERS AND ADDITIVES, BLENDS OF THE ELASTOMERS WITH OTHER POLYMERS, CHEMICAL RESISTANCE, AGEING, FLAMMABILITY, GAS PERMEABILITY, BLOCK COPOLYMERS, AND THERMOPLASTIC ELASTOMERS BASED ON ETHYLENE-PROPYLENE, AND APPLICATIONS FOR ETHYLENE PROPYLENE ELASTOMERS, E.G. BUILDING APPLICATIONS, SEALING STRIPS, SPORTS SURFACES, ELECTRIC CABLES, PROTECTIVE COATINGS AND ADHESIVES. A LIST OF PATENTS AND DETAILS OF FORMULATIONS ARE GIVEN. 863 REFS.

OPPORTUNITIES IN CRIENTED NITRILE RESIN CONTAINERS 80-09 52281L

CARLSON. J. A. BORLA, L.

MOD.PLAST.INT.,10,NO.6,JUNE 1980,P.38-40

CONSIDERATION IS GIVEN TO THE PRODUCTION OF BOTTLES AND OTHER CONTAINERS FOR BEVERAGES, FOODSTUFFS, ETC., BY ORIENTATION BLOW MOULDING OF ACRYLONITRILE COPOLYMERS, I.E. COPOLYMERS OF ABOUT 70% ACRYLONITRILE WITH ACRYLATES OR STYRENE, POSSIBLY CONTAINING ELASTOMERIC IMPACT MODIFIERS. THE ACRYLONITRILE MONOMER PROVIDES HIGH TENSILE STRENGTH, STIFFNESS, GAS BARRIER AND CHEMICAL RESISTANCE PROPERTIES, AND THE COMONOMER CONTRIBUTES PROCESSIBILITY. DATA ARE PRESENTED ON TEMPERATURE AND DRAW RATIO FACTORS WHICH CONTROL ENHANCEMENT OF MECHANICAL PROPERTIES AND PERMEABILITY CHARACTERISTICS DURING ORIENTATION. COMMERCIALLY AVAILABLE COPOLYMERS ARE BAREX 210 (VISTRON CORP.) AND SOLTAN (SOLVAY & CIE.). DEVELOPMENTS IN PROCESSING METHODS AND MARKET POTENTIALS FOR SUCH CONTAINERS ARE SURVEYED.

GAS EDUS DIFFUSION AND PERMEATION IN CLOSED-CELL POLYMERIC FDAMS 80-09 52620C

MEHTA, B. S. COLOMBO, E. A.

WASHINGTON, D.C., APRIL 1978, P.689-92. CONFER. 012 PUBLCN. DETAILS- SOC. PLAST. ENGRS. 36TH ANNUAL TECHNICAL CONFERENCE

A QUANTITATIVE EXPLANATION IS PRESENTED OF THE EFFECT OF TEMP., BLOWING GAS CONCENTRATION AND STRUCTURAL CHARACTERISTICS, SUCH AS CELL SIZE AND CELL ORIENTATION, ON GASEOUS DIFFUSION AND PERMEATION OF CLOSED-CELL PS FOAM. 8 REFS. 42C21-6124-93513

GAS PERMEABILITY OF POLYMERIC MEMBRANES 80-08 51719L

BELEN KAYA, N. M. BUBLEVSKII, I. M. GARNIZOVA, A. N.

PLAST.MASSY, NO.10, 1979, P.60

RUSSIAN. 2 REFS. ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

PERMEATION OF GASES AND VAPOURS THROUGH POLYMER FILMS AND THIN SHEET. I. 80-08 51211L

LOMAX, M.

POLYM.TEST.,1,NO.2,APRIL/JUNE 1980,P.105-47

A REVIEW IS MADE OF TECHNIQUES FOR MEASURING THE PERMEATION OF GASES AND VAPOURS THROUGH POLYMER FILMS AND THIN SHEETS. 67 REFS.

EFFECT OF VARIOUS FACTORS ON GAS PERMEABILITY OF POLYMERIC FILMS. I. AIR PERMEABILITY OF POLYSTYRENE FILMS 80-07 50126L

KAMINSKA, A.

POLIM.TWORZ.WEILK.,25,NO.2,FEB.1980,P.47-8

POLISH. THE PERMEABILITY TO AIR OF PS FILMS PREPARED BY CASTING POLYMER SCLUTIONS ON GLASS PLATES AND EVAPORATING THE SOLVENT WAS DETERMINED. FILMS WERE PREPARED FROM PS FRACTIONS OF DIFFERING MOLEC.WT. AND FROM MIXTURES OF FRACTIONS EXHIBITING SIMILAR INTRINSIC VISCOSITY BUT DIFFERENT POLYDISPERSITY. AIR PERMEABILITY WAS FOUND TO INCREASE WITH INCREASE IN MOLEC.WT. AND WITH BROADENING OF MWD. 18 REFS. ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

GAS PERMEABILITY OF METALLISED POLYMER FILMS 80-06 48608L

KAPANIN, V. V. REITTLINGER, S. A. STRUNINA, O. B. PRILIPOV, V. V.

VYS.SOED.B,19,NO.12,DEC.1977,P.903-8

RUSSIAN. THE GAS PERMEABILITY OF A LAMINATE CONSISTING OF PE FILM SANDWICHED BETWEEN, ON EACH SIDE, A LAYER OF ALUMINIUM AND A LAYER OF PETP WAS INVESTIGATED AND PROMISING RESULTS OBTAINED. 5 REFS.

DETERMINATION OF THE SOLUBILITY OF GASES IN POLYMER FILMS BY GAS CHROMATOGRAPHY 80-06 48962L

KAPANIN, V. V. SIROTIN, Y. A. D. REITLINGER, S. A. PRILIPOV, V. V.

POLYM.SCI.USSR,21,NO.2,1979,P.500-2

SOLUBILITY COEFFICIENTS ARE DETERMINED FOR HELIUM, DXYGEN, ARGON AND NITROGEN IN LDPE FILMS. THESE RESULTS SHOWED SATISFACTORY AGREEMENT WITH RESULTS IN THE LITERATURE. 9 REFS.

INFLUENCE OF ADDITION OF OLIGOMER IC VINYLALKYLDIMETHYL SILANES ON THE GAS PERMEABILITY OF POLYVINYL TRIMETHYLSILANE 80-06 49196L

EVSEENKO, A. L. TEPLYAKOV, V. V. DURGAR'YAN, S. G. NAMETKIN, N. S.

VYS.SOED.B.21.NO.2.1979.P.153-5

RUSSIAN. 5 REFS. ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

IRRADIATION EFFECT ON GAS DIFFUSION IN POLYMER FILMS. I. HYDROGEN DIFFUSION THROUGH MYLAR FILM 80-06 47953C

RAO, K. A. PUSHPA, K. K. IYER, R. M.

GUJARAT, FEB. 1979, PP.1. CONFER. 012 PUBLON. DETAILS-INDIA, BOARD OF RES. NUCLEAR SCI. INDUS. POLYM. & RADIAT. SYMPOS.

(ABSTRACT ONLY). 43C112-625-93513

ISOTOPE EFFECT IN THE DIFFUSION OF HYDROGEN AND DEUTERIUM IN POLYMERS 80-06 48299L

TOI. K. TAKEUCHI, K. TOKUDA, T.

J.POLYM.SCI.POLYM.PHYS., 18, NO. 2, FEB. 1980, P. 189-98

TEMPERATURE DEPENDENCES OF DIFFUSION AND PERMEATION COEFFICIENTS OF HYDROGEN AND DEUTERIUM IN GLASSY AND RUBBERY POLYMER FILMS WERE MEASURED. THE SIZE OF THE FREE VOLUME ELEMENT IN RUBBERY POLYMERS WAS CALCULATED ACCORDING TO THE THEORY OF FRISCH AND ROGERS FOR THE QUANTUM ISOTOPE EFFECT, BUT THE FREE VOLUME WAS TOO LARGE FOR PRECISE CALCULATION BELOW THE TG. THE COOPERATIVE MOVEMENT OF SEGMENTS IS ALSO DISCUSSED USING THE RATIO OF PRE-EXPONENTIAL FACTORS FOR DIFFUSION MECHANISMS ABOVE AND BELOW THE TG. 7 REFS. 93513

ENTROPY CORRELATION THEORY AND DIFFUSION MEASUREMENTS FOR ORIENTED POLYMERS 80-04 45453L

BARKER. R. E. TSAI, R. C. WILLENCY, R. A.

J.POLYM.SCI.POLYM.SYMP.,NO.63,1978,P.109-29

THE ENTROPY CORRELATION THEORY PREDICTS THAT ANY STRUCTURAL CHANGE WHICH DECREASES THE CONFIGURATIONAL ENTROPY OF A COMPLEX MOLECULAR SYSTEM WILL LEAD TO AN APPROXIMATELY EQUAL INCREASE IN THE ACTIVATION ENTROPY FOR STRUCTURALLY CONTROLLED RATE PROCESSES SUCH AS DIFFUSION. CONFIRMATION OF THE THEORY IS PRESENTED FOR THE CASE OF GAS DIFFUSION IN PERMANENTLY STRETCHED POLYMERIC MEMBRANES. SYSTEMS

INVESTIGATED INCLUDED REPRESENTATIVE POLYALKYL METHACRYLATES AND A SELECTION OF NON-POLAR AND POLAR GASES. 35 REFS. 93513

ENERGETICS OF GAS SORPTION IN GLASSY POLYMERS 79-12 39757L

KOROS, W. J. PALL, D. R. HUVARD, G. S.

POLYMER, 20, NO.8, ALG. 1979, P. 956-60

AN ANALYTICAL EXPRESSION WAS DERIVED FOR PREDICTION OF THE MOLAR ENTHALPY OF SORPTION AT CONSTANT CONCENTRATION IN DUAL MODE SORPTION SYSTEMS. THE EXPRESSION WAS SHOWN TO PROVIDE A GOOD DESCRIPTION OF DATA DERIVED BY FORMAL THERMODYNAMIC ANALYSIS OF CARBON DIOXIDE SORPTION ISOTHERMS IN PETP. 13 REFS. 93513

PERMEABILITY OF METALLISED POLYMER FILMS. I. METHOD FOR GAS PERMEABILITY MEASUREMENT 79-12 39758L

SPRINGER, J. BRITO, H.

J.APPL.POLYM.SCI.,24,NO.2,15TH JULY 1979,P.329-37

GERMAN. THE CONSTRUCTION AND OPERATION OF AN APPARATUS FOR THE DETERMINATION OF GAS PERMEABILITY THROUGH METALLISED POLYMER FILMS IS DESCRIBED. THE TEST GASES (NITROGEN, DXYGEN AND CARBON DIOXIDE) PENETRATE UNDER PRESSURE DIFFERENCES FROM 100 TORR TO 20 BAR THROUGH METALLISED ABS FILMS. THE METAL LAYERS CONSIST OF CHEMICALLY DEPOSITED NICKEL AND GALVANIC DEPOSITED COPPER. THE QUANTITY OF PERMEATED GASES IS DETERMINED BY GAS CHROMATOGRAPHY, AND LEAK EFFECTS CAN BE MEASURED QUANTITATIVELY. THE PERMEABILITY OF GAS MIXTURES, I.E. AIR, CAN ALSO BE INVESTIGATED. THE APPARATUS ALLOWS THE DETERMINATION OF EXTREMELY LOW PERMEABILITY RATES AS WELL AS THOSE FOR CONVENTIONAL POLYMER SYSTEMS. 29 REFS. 293513T

PROPERTIES AND APPLICATIONS OF POLYMER ALLOYS 79-11 38007L

GRECO. R. MARTUSCELLI, E.

CHIMICA E INC., 61, NO. 4, APRIL 1979, P. 298-309.

ITALIAN. THE CLASS OF MATERIALS CONSIDERED INCLUDES POLYMER BLENDS, GRAFT COPCLYMERS, BLOCK COPOLYMERS AND INTERPENETRATING POLYMER NETWORKS. AN EXAMINATION IS MADE OF THERMODYNAMIC COMPATIBILITY, HOMOGENEITY, ELASTIC MODULUS, RUPTURE RESISTANCE, IMPACT PROPERTIES, ABRASION RESISTANCE, THERMAL PROPERTIES, THERMAL AND LIGHT DEGRADATION, FLAMMABILITY, GAS AND VAPOUR PERMEABILITY OF FILMS AND MEMBRANES, CRYSTALLISATION KINETICS, RECYCLING, AND COUPLING AGENTS. MAJOR PROBLEMS TO BE SOLVED INCLUDE HETEROGENEITY AND INCOMPATIBILITY, AND COUPLING AGENTS AND/OR INTERFACIAL PROPERTY MODIFIERS NEED TO BE DEVELOPED. 36 REFS. 6125

WEATHERING PROPERTIES OF POLYMERS IN CABLES 79-11 38147L

HULE MEX.PLAST., 34,NO.397,FEB.1979,P.5/20

CHEVASSUS, F.

SPANISH. A DETAILED STUDY IS MADE OF THE EFFECTS OF WEATHERING ON RUBBER AND PLASTICS CABLE INSULATION, WITH REFERENCE TO CABLES OF ALL TYPES. THE ACTION OF LIGHT AND OF LIGHT STABILISERS, HUMIDITY, OZONE AND OTHER GASES IS CONSIDERED FOR DIFFERENT POLYMERS, AND ELECTRICAL PROPERTIES ARE DISCUSSED. 6E1-93

EVAPORATION OF POLYMER- SOLVENT MIXTURES. DETERMINATION OF VAPOUR PRESSURES FROM GAS EOUS DIFFUSION COEFFICIENTS 79-10 37611L

COCA, J. BUENO, J. L. ALVAREZ, R.

POLYM.BULL.,1,NO.7,MAY 1979,P.459-64

THE STEFAN-WINKELMANN DIFFUSION TECHNIQUE WAS USED TO DETERMINE VAPOUR PRESSURES OF HIGH BOILING POINT COMPOUND/SOLVENT MIXTURES. DATA ARE REPORTED AT A TEMP. OF 67C FOR THE MIXTURES POLYPHENYL ETHER (6 RINGS)/BENZENE AND CARBOWAX 1500/BENZENE AND AT 100C FOR POLYPHENYL ETHER/TOLUENE AND TRICRESYL PHOSPHATE/TOLUENE. THE RANGE OF CONCENTRATION WAS ONLY LIMITED BY THE APPEARANCE OF A SOLID PHASE AND RESULTS WERE IN GOOD AGREEMENT WITH THOSE DETERMINED BY VAPOUR-PRESSURE OSMOMETRY. 6 REFS. 43C52-93513T

EVALUATION OF RESULTS OF SIMPLIFIED DETERMINATION METHODS OF POLYMER MEMBRANE PERMEABILITY TO VAPOURS 79-10 36773L

IZYDORCZYK, J. FODKOWKA, J. SALWINSKI, J.

J.APPL.POLYM.SCI.,23,NO.8,15TH APRIL 1979,P.2265-9

PERMEABILITY COEFFICIENTS FOR SOME SELECTED SYSTEMS COMPRISING A POLYMER MEMBRANE AND ORGANIC VAPOURS WERE MEASURED BY SIMPLIFIED METHODS, WITH THE AIM OF EVALUATING THE SUITABILITY OF THESE TECHNIQUES FOR MEMBRANE PERMEABILITY DETERMINATION UNDER AVERAGE CONDITIONS OF USE. RESULTS OBTAINED BY WEIGHED CELL, CAPILLARY EVAPORATION AND ELECTROCHEMICAL METHODS WERE COMPARED AND ANALYSED, TAKING INTO ACCOUNT PERMEATION MODELS ASSOCIATED WITH DIFFERENT APPARATUS AND OPERATION PRINCIPLES AS WELL AS DIFFERENT MEASURING CONDITIONS. FOR SIMILAR MASS TRANSFER MODELS, PERMEABILITY COEFFICIENT VALUES OF THE SAME ORDER AND CLOSE ACCURACY OF MEASUREMENT WERE OBTAINED. 8 REFS. 6M-93513

PERMEABILITY OF DXYGEN THROUGH POLYMERS. I. A NOVEL SPECTROPHOTOCHEMICAL METHOD 79-10 36776L

PETRAK, K.

J.APPL.POLYM.SCI.,23,NO.8,15TH APRIL 1979,P.2365-71

A NEW METHOD FOR THE MEASUREMENT OF DXYGEN PERMEABILITY THROUGH POLYMER MEMBRANES IS BASED ON MONITORING THE SENSITISED PHOTO-CXYGENATION OF A SINGLET DXYGEN ACCEPTOR IN A DETECTOR LAYER SANDWICHED BETWEEN A SUPPORT AND THE POLYMER LAYER. THE DETECTOR LAYER CONTAINS A SENSITISER WHICH ON IRRADIATION PRODUCES SINGLET EXCITED DXYGEN FROM THE GROUND-STATE CXYGEN AVAILABLE. THE SINGLET DXYGEN REACTS WITH AN OXYGEN ACCEPTOR, THE DISAPPEARANCE OF WHICH CAN BE FOLLOWED BY SPECTROPHOTOMETRY. IN THE PHOTOSTATIONARY STATE, CHANGES IN ACCEPTOR ABSORBANCE ARE DIRECTLY RELATED TO THE OVERALL FLUX OF DXYGEN THROUGH THE MEMBRANE. PERMEATION COEFFICIENT OF DXYGEN IS PROPORTIONAL TO THE RATE OF CHANGE IN ACCEPTOR ABSORBANCE AND TO THE INVERSE OF THE DAYGEN CONCENTRATION IN THE SURROUNDING ATMOSPHERE. DXYGEN PERMEABILITY WAS MEASURED FOR A GROUP OF WATER-SOLUBLE POLYMERS. 13 REFS. 6M-93513T

INFLUENCE OF THE STRUCTURAL ISOMERISM OF A POLYMER ON THE PERMEABILITY DIFFLSION AND SOLUBILITY OF HELIUM, HYDROGEN, NEON AND ACETONE VAPOUR 79-10 36895L

TIKHOMIROV, B. P. POLYAK, M. A.

IZV.VUZ KH.I KH.TEKH.,22,NO.2,1979,P.211-4

RUSSIAN. A STUDY IS DESCRIBED OF THE PERMEABILITY OF POLYMETHYL ACRYLATE (PMA) TO HELIUM, HYDROGEN AND NEON, AND THE DIFFUSION AND SOLUBILITY COEFFICIENTS IN THE GLASS TRANSITION TEMPERATURE RANGE WERE DETERMINED. THE ACTIVATION ENERGY OF DIFFUSION OF ACETONE INTO PMA, AND THE COEFFICIENT OF DIFFUSION, ARE COMPARED WITH THOSE FOR PVAC, AND REASONS FOR THE DIFFERENCE ARE ADVANCED. 5 REFS. 42C35111-93513 ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

GAS-PERMEABILITY OF POLYMERIC PACKAGING FILM 79-08 35301L

DODONOV, A. M. MURAVIN, YA. G.

PLAST.MASSY, NO.12, 1978, P.53

RUSSIAN. DATA ARE PRESENTED ON THE GAS PERMEABILITY OF PACKAGING MATERIALS, TOGETHER WITH A DESCRIPTION OF A DEVICE USED FOR TESTING THIS PROPERTY. 2 REFS. 6P11-93513 ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

DIFFUSION OF GASES THROUGH POLYURETHANE BLOCK POLYMERS 79-08 35444L

MCBRIDE, J. S. MASSARO, T. A. COOPER, S. L.

J.APPL.POLYM.SCI.,23,NO.1,1ST JAN.1979,P.201-14

THE DIFFUSIVITIES OF SIMPLE GASES THROUGH A SERIES OF PUBLOCK COPOLYMERS OF DIFFERING AROMATIC URETHANE CONTENT AND TYPE OF SOFT SEGMENT WERE MEASURED USING A QUADRAPOLE MASS SPECTROMETER AS A DETECTING DEVICE. ALTHOUGH AN ARRHENIUS EXPRESSION GENERALLY DESCRIBED THE TEMP. DEPENDENCE OF DIFFUSION IN THIS SYSTEM, A DISCONTINUITY WAS OBSERVED IN THE ARRHENIUS PLOTS FOR SOME MATERIALS AND THE DISCONTINUITY WAS FOUND TO BE RELATED TO THE ONSET OF THE TG IN THE HARD DOMAINS. POLYESTER-URETHANES HAD LOWER ACTIVATION ENERGIES FOR DIFFUSION THAN POLYETHER-URETHANES OF SIMILAR HARD SEGMENT COMPOSITION. 36 REFS. 43C6-93512

DIFFUSION CELL FOR THE STUDY OF GAS TRANSFER THROUGH REINFORCED PCLYMER MATERIALS 79-06 32784L

KAPANIN, V. V. PRILIPOV, V. V.

POLYM.SCI.USSR,19,NO.5,1977,P.1345-8

A DIFFUSION CELL WAS DESIGNED TO EXAMINE GAS TRANSFER

THROUGH REINFORCED FILM MATERIALS OVER A WIDE RANGE OF TEMPERATURES. THE GAS PERMEABILITY OF RUBBERISED FABRICS OF TERYLENE, LAVSAN (PETP) AND COTTON WAS STUDIED IN THE TEMP RANGE 20-80C. IT WAS SHOWN THAT THE LOGARITHMIC DEPENDENCE OF PERMEABILITY ON INVERSE TEMP. IS NON-LINEAR. THE EFFECTIVE ACTIVATION ENERGY OF PERMEABILITY WAS DETERMINED FOR THE TEMP. RANGE OF 20-40C. 3 REFS. 628-93513T

BARRIER POLYMERS 79-06 32280C

SALAME, M. STEINGISER, S.

NEW YORK, APRIL 1976, P.18-28. CONFER. 012 PUBLCN. DETAILS-ACS, CHEM. MARKET. & ECON.DIV. HISTORICAL... CHEM. MARKET... SYMPOSIA

THE PERMEABILITY OF POLYMERS AS DETERMINED BY STRUCTURAL AND MORPHOLOGICAL PREPERTIES OF BOTH THE POLYMER MATRIX AND THE PERMEATING SPECIES IS DISCUSSED. THE CHARACTERISTICS OF HIGH BARRIER POLYMERS ARE RESISTANCE TO GAS FLOW (OF OXYGEN, CARBON DIOXIDE AND NITROGEN) LIQUID (WATER) FLOW, AND RESISTANCE TO ABSCRPTION OF ORGANIC MOLECULES. THE CALCULATION AND USE OF PERMACHLOR VALUES IN PREDICTING PERMEABILITY AND THE BARRIER PROPERTIES AND APPLICATIONS OF HIGH BARRIER POLYMERS E.G. SAN ARE CESCRIBED. DETAILS OF OTHER POLYMERS ARE PRESENTED IN TABULAR FORM. 12 REFS. 6L-9351

APPARATUS FOR STUDYING THE GAS PERMEABILITY OF POLYMERS 79-02 28108L

SILONOV, YU. A. KOLESNIKOV, A. N.

PLAST.MASSY, NO.8, 1978, P.67-8

RUSSIAN. A LABORATORY APPARATUS FOR TESTING THE GAS PERMEABILITY OF PCLYMERIC MATERIALS UNDER VACUUM AND HIGH PRESSURE AT VARIOUS TEMPERATURES IS DESCRIBED. 2 REFS. 293513T ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

DETERMINATION OF PERMEABILITY OF POLYMERIC MEMBRANES 78-12 24850L

TEPLYAKOVA, V. V. EVSENKO, A. L. NOVITSKII, E. G. DURGAR'YAN, S. G.

PLAST.MASSY,NO.5,1978,P.49-51

RUSSIAN. METHODS ARE PRESENTED FOR DETERMINING THE GAS PERMEABILITY OF FCLYMERIC MEMBRANES, INCLUDING INTEGRAL AND DIFFERENTIAL VARIANTS OF MEASUREMENT OF THE COEFFICIENTS OF PERMEABILITY AND CIFFUSION OF GASES IN POLYMERS, USING GAS CHROMATOGRAPHS OVER A WIDE RANGE OF TEMPERATURES, PRESSURES AND STREAMS OF PENETRANT. THE COEFFICIENTS OF PERMEABILITY AND DIFFUSION OF VARIOUS GASES IN POLYVINYLTRIMETHYLSILOXANE ARE GIVEN. 12 REFS. 6M-93513T ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

NOVEL TECHNIQUE FCR MEASURING THE DIFFUSION CONSTANT OF DXYGEN IN POLYMER FILMS 79-01 26081L

MACCALLUM, J. R. RUDKIN, A. L.

EUR.POLYM.J.,14,NC.9,1978,P.655-6

A SIMPLE TECHNIQUE IS DESCRIBED FOR THE MEASUREMENT OF THE DIFFUSION CONSTANT FOR OXYGEN IN POLYMERIC GLASSES, AND MEASURED VALUES FOR PS AND PMMA ARE REPORTED. MEASUREMENTS MAY BE MADE OVER A WIDE RANGE OF TEMPERATURES. 3 REFS. 93513T

CHARACTERISATION CF POROUS POLYMERIC MEMBRANES BY GAS PERMEABILITY 79-01 26459L

NOHMI, T. MANABE, S. KAMIDE, K. KAWAI, T.

KOBUNSHI RONBUN.,35,NO.8,AUG.1978,P.509-16

JAPANESE. AN ATTEMPT WAS MADE TO ESTABLISH A METHOD FOR ESTIMATING THIRD AND FOURTH AVERAGE PORE RADII AND PORE SIZE

FREQUENCY DISTRIBUTION FROM EXPERIMENTAL DATA ON GAS PERMEABILITY COEFFICIENT OF POROUS MEMBRANES. 7 REFS. 6M-93513

INTRODUCTION OF PLASTICS IN BIOGAS 79-01 26881L

KALIA, A. K. SINGH, R. B.

POP.PLAST., 23, NO.4, APRIL 1978, P.52-5

THE FINDINGS ARE FRESENTED OF TESTS INTO THE SUITABILITY OF POLYMERS SUCH AS PVC, SYNTHETIC RUBBER AND NYLON FOR THE STORAGE AND TRANSPORTATION OF GAS DERIVED FROM DUNG AND OTHER ORGANIC WASTES. RIGID PVC SHEET WAS FOUND TO BE SUITABLE AS A GAS HOLDER, AND PVC/NYLON LAMINATED SHEET WORKED SATISFACTORILY FOR THE TRANSFORTATION OF THE GAS. 1 REF. 63AG

PROBLEM OF DETERMINATION OF THE CONCENTRATION DEPENDENCE OF THE COEFFICIENT OF ACETONE DIFFUSION IN POLYMERS 79-01 27117L

BELYAYEV, O. F. VOYEVODSKII, V. S. BEZRUKAVNIKOVA, L. M. MAIZELIS, B. A.

POLYM.SCI.USSR, 18, NO.6, 1976, P. 1543-7

A STUDY WAS MADE OF LATEX FILMS BASED ON POLYISOPRENE, POLYCHLOROPRENE AND NITRILE RUBBERS TO DETERMINE THEIR PERMEABILITY TO ACETONE VAPOUR. DIFFUSION COEFFICIENTS WERE OBTAINED FOR DIFFERENT SOLVENT CONCENTRATIONS USING A PROPOSED METHOD FOR CALCULATING THE CONCENTRATION DEPENDENCE OF THE COEFFICIENTS. 8 REFS. 93513

METHOD OF MEASURING THE DIFFUSION AND SOLUBILITY OF GASES IN POLYMERS 79-C1 27119L

SAPOZHNIKOV, D. N. SHLYAKHOV, R. A. TOCHIN, V. A.

PLAST.MASSY, NO.6, 1978, P.68-70

RUSSIAN. 8 REFS. 93513T ARTICLES FROM THIS JOURNAL CAN BE REQUESTED FOR TRANSLATION BY SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

LIQUID POLYSULPHIDE POLYMERS 78-09 21549L

WILHELM. G.

ADHASION, 22, NO.5, MAY 1978, P.156-60

GERMAN. THE AUTHOR REVIEWS CHEMICAL COMPOSITION AND PROPERTIES OF THE THIOKOL RANGE OF LIQUID POLYSULPHIDE POLYMERS. DIFFERENCES IN CROSSLINK DENSITY RESULT IN POLYMERS OF DIFFERENT VISCOSITY, BUT ALL CONTAIN MERCAPTO AND GROUPS WHICH PERMIT CURING BY ATMOSPHERIC DXIDATION OR BY OXIDATION BY MEANS OF METAL DXIDES OR ORGANIC PEROXIDES. CURE RATE CAN BE CONTROLLED CHEMICALLY BY MEANS OF ADDITIVES AND THE POLYMERS CAN BE USED TOGETHER WITH PHENOL-FORMALDEHYCE, POLYESTER OR EPOXY RESINS. TOXICOLOGICAL PROPERTIES, STRENGTH, ELECTRICAL PROPERTIES, GAS AND WATER PERMEABILITY, ADHESIVE PROPERTIES AND THERMAL AND SOLVENT STABILITY ARE DISCUSSED, AND A RANGE OF APPLICATIONS REVIEWED. 32 REFS. 43C52-8962

MECHANISM OF GAS FERMEATION THROUGH POROUS POLYMERIC MEMBRANES 78-C9 20897L

NOHMI. T. MAKINO. H. MANABE. S. KAMIDE. K. KAWAI. T.

KOBUNSHI RONBUN.,35,NO.4,APRIL 1978,P.253-61

JAPANESE. THE EFFECTS OF PORE SIZE DISTRIBUTION AND THE CHEMICAL NATURE OF THE PERMEATING GAS ON THE PERMEABILITY COEFFICIENT WERE INVESTIGATED. GAS FLOW WAS FOUND TO

CONSIST OF A MIXTURE OF VISCOUS AND SLIP FLOW. THE DEPENDENCE OF THE PERMEABILITY COEFFICIENT OF INORGANIC GASES ON PORE SIZE DISTRIBUTION AND THE MEAN FREE PATH OF THE GAS AGREED WITH A PREVIOUSLY PROPOSED THEORETICAL EQUATION WHEREAS THOSE OF ORGANIC GASES WERE MUCH LARGER THAN THE THEORETICAL VALUES. IT WAS SUGGESTED THAT INTERACTION BETWEEN MEMBRANE AND GAS COULD BE THE CAUSE OF OBSERVED DIFFERENCES FOR ORGANIC GASES. 14 REFS. 6M-93513

PARYLENE POLYMERS. I. SYNTHESIS, PROPERTIES AND IMPORTANCE 78-08 20094L

BALDOUF, L. HAMANN, C. LIBERA, L.

PLASTE U.KAUT., 25, NO. 2, FEB. 1978, P. 61-4

GERMAN. THE PREPARATION OF POLY-P-XYLYLENES (PARYLENES) IS DESCRIBED AND THEIR DEPOSITIONS AS FILMS ON COOLED SURFACES BY BI-RADICAL COMBINATION METHODS, VIA THE PYROLYSIS OF P-XYLENE AND DI-P-XYLENE DERIVATIVES IN LOW-PRESSURE SYSTEMS, IS DISCUSSED. EXTREMELY UNIFORM POLYMER COATINGS OF THICKNESS BOA TO 100 MICRONS CAN BE ACHIEVED ON METAL, GLASS, WOOD OR PAPER, THEIR PURITY, MOLECULAR WEIGHT AND TRANSPARANCY BEING DETERMINED BY THE NATURE OF THE STARTING MATERIALS AND DEPOSITION CONDITIONS. ELECTRICAL AND MECHANICAL PROPERTIES AND GAS PERMEABILITY ARE EXAMINED AND USE OF THE FILMS AS DIELECTRICS IS DISCUSSED. 60 REFS. 42W

GAS PERMEATION OF POLYMER BLENDS. V. COMPATIBILITY STUDIES OF POLY (VINYL CHLORIDE) /POLY-EPSILON-CAPROLACTONE BLENDS 78-08 19713L

SHUR, Y. J. RANBY, B.

J.MACROMOL.SCI.B, 14, NO.4, 1977, P.565-72

THE TRANSPORT BEHAVIOUR OF OXYGEN AND NITROGEN AT 25C WAS STUDIED FOR PHYSICAL BLENDS OF PVC/PCL AND FOUR TYPES OF ETHYLENE-VINYL ACETATE AND ACRYLONITRILE-BUTADIENE COPOLYMERS. THE FVC/PCL BLENDS WERE SHOWN TO FORM A COMPATIBLE SYSTEM. IT WAS ALSO SHOWN THAT AT PCL CONTENTS OF LESS THAN 30 kT.%, A SEPARATE CRYSTALLISED PCL PHASE WAS FORMED. THE RESULTS ARE DISCUSSED IN TERMS OF COOPERATIVE

SEGMENTAL RELAXATION PROCESSES BETWEEN PVC AND PCL CHAINS. 14 REFS. 42C382-6125-93513

HIGH DENSITY POLYETHYLENE 78-06 16996L

ANON

MAT.PLAST.ELAST., NO.1, JAN. 1978, P. 34-7

ITALIAN. CONSIDERATION IS GIVEN TO THE PRODUCTION OF HOPE WITH REFERENCE TO COMMONLY USED POLYMERISATION PROCESSES, AND TO THE PROPERTIES OF THIS MATERIAL, I.E. CHEMICAL PROPERTIES, GAS PERMEABILITY, MECHANICAL, ELECTRICAL AND THERMAL PROPERTIES. DETAILS ARE ALSO GIVEN OF COMPOUNDING INGREDIENTS, PROCESSING AND APPLICATIONS, AND PROPERTIES OF HDPE AND COMPARED WITH THOSE OF LDPE AND PP. A LIST OF HDPE TRADE MARKS AND MANUFACTURERS IS INCLUDED. 42C11

ESTIMATION OF POLYMER SOLUBILITY PARAMETERS BY GAS 78-07 18239L CHROMATOGRAPHY

DIPAULA-BARANYI, G. GUILLET, J. E.

MACROMOLECULES, 11, NO.1, JAN. / FEB. 1978, P. 228-35

PARTIAL MOLAR HEATS OF MIXING, PARTIAL MOLAR FREE ENERGIES OF MIXING AND THE FLORY-HUGGINS CONSTANT WERE DETERMINED FOR A VARIETY OF HYDRCCARBONS IN PS AND POLYMETHYL ACRYLATE BY GAS CHROMATOGRAPHY. SOLUBILITY COEFFICIENTS AT INFINITE DILUTION WERE CALCULATED FROM THE THERMODYNAMIC DATA AND COMPARED WITH THEORETICAL PREDICTIONS. 33 REFS. 42C21-9351 SOME ASPECTS OF PLASMA COPOLYMERISATION OF ACETYLENE WITH NITROGEN AND/OR WATER 78-06 16346L

YASUDA, H. T HIROTSU

J.POLYM.SCI.POLYM.CHEM., 15, NO.11, NOV. 1977, P.2749-71

PLASMA POLYMERISATIONS OF MIXTURES OF ACETYLENE-NITROGEN, ACETYLENE-WATER AND ACETYLENE-NITROGEN-WATER WERE INVESTIGATED USING AN ELECTRODELESS GLOW DISCHARGE FROM A 13.5 MHZ RADIOFREQUENCY SOURCE. PROPERTIES OF PLASMA POLYMERS WERE EXAMINED AS FUNCTIONS OF MOLE RATIOS OF NITROGEN AND/OR WATER TO ACETYLENE. PROPERTIES INVESTIGATED INCLUDE INTERNAL STRESS, GAS PERMEABILITY AND SURFACE ENERGY. 24 REFS. 42F1A-7223

POLYMERISATION OF ORGANIC COMPOUNDS IN AN ELECTRODELESS GLOW DISCHARGE. IX 78-05 15701L

YASUDA, H. HIRETSU, T.

J.APPL.POLYM.SCI.,21,NO.11,NOV.1977,P.3167-77

PROPERTIES (FREE-RADICAL CONCENTRATION, GAS PERMEABILITIES, INTERNAL STRESS, AND CONTACT ANGLE OF WATER) OF PLASMA POLYMERS OF ACETYLENE AND OF ACRYLCNITRILE WERE INVESTIGATED AS A FUNCTION OF FLOW RATE OF MONOMER. IT WAS FOUND THAT THE FLOW RATE HAS A STRONG INFLUENCE ON FREE-RADICAL CONCENTRATION, GAS PERMEABILITIES AND INTERNAL STRESS BUT LITTLE INFLUENCE ON THE CONTACT ANGLE OF WATER. THE DISCHARGE POWER HAS LITTLE EFFECT ON PROPERTIES WHEN FULL GLOW IS MAINTAINED. GAS PERMEABILITIES DECREASE WITH INCREASING CONCENTRATION OF FREE RADICALS. 11 REFS. (PT.VIII, IBID. P.3139-46) 42F1-7223

NOBLE GAS PERMEABILITY OF POLYMER FILMS AND COATINGS 78-03 12274L

HAMMON, H. G. ERNST, K. NEWTON, J. C.

J.APPL.POLYM.SCI.,21,NO.7,JULY 1977,P.1989-97

PERMEABILITIES OF NOBLE GASES, PARTICULARLY ARGON, KRYPTON AND XENON, WERE MEASURED THROUGH A NUMBER OF POLYMER FILMS AND COATINGS. EXTRAPOLATION OF THE LOG OF THE PERMEATION COEFFICIENT VERSUS THE SQUARE OF THE GAS MOLECULAR DIAMETER WAS USED TO ESTIMATE RANDOM PERMEABILITY. AN EQUATION IS PRESENTED FOR PRECICTING PERMEABILITY TO THESE NOBLE GASES AS A FUNCTION OF THE BASE POLYMER STRUCTURE OF THE COATING. 14 REFS. 625-93513

DETERMINATION OF DIFFUSION COEFFICIENTS IN POLYETHYLENE BY GAS CHROMATOGRAPHY 78-03 12526L

MILLEN, W. HAWKES, S. J.

J.POLYM.SCI.POLYM.LETT., 15, NO.8, AUG. 1977, P.463-5

AN EXPRESSION FOR CALCULATING THE STATIONARY PHASE MASS TRANSFER TERM FOR THE DIFFUSIVITY OF ORGANIC MOLECULES IN LIQUID POLYMERS IS DISCUSSED. REVISED VALUES OF DIFFUSION COEFFICIENTS FOR SEVERAL SOLUTES IN LOPE USING THIS EQUATION ARE TABULATED. 8 REFS. 42C11-9351

DIFFUSION CELL FOR STUDY OF GAS TRANSFER THROUGH WALLS OF POLYMER TUBES 78-02 11562L

KAPANIN, V. V. PRILIPOV, V. V.

POLYM.SCI.USSR, 18, NO. 3, 1976, P.820-3

THE DESIGN OF A DIFFUSION CELL FOR INVESTIGATING THE TRANSFER OF LOW MCLEC.WT. SUBSTANCES THROUGH POLYMER TUBE WALLS, USING A GAS CHROMATOGRAPH, IS DESCRIBED. USING LOPE TUBES, IT WAS SHOWN THAT PRESSURE HAD NO EFFECT ON THE PERMEABILITY FACTER AND THAT THE TEMP. DEPENDENCE OF THE

PERMEABILITY FACTOR WAS SIMILAR TO THAT OF FILMS. 8 REFS. 293513T

BARRIER POLYMERS 78-01 11207L

SALAME, M. STEINGISER, S.

POLYM.PLAST.TECHNEL.ENGNG.,8,NO.2,1977,P.155-75

FACTORS AFFECTING POLYMER BARRIER PROPERTIES AND THE EFFECTS OF POLYMER PARAMETERS, TEMP. AND THICKNESS AND PERMEATING SPECIES ON POLYMER BARRIER PROPERTIES ARE DISCUSSED. THE PREDICTION OF PERMEABILITY, DEFINITION AND CLASSIFICATION OF HIGH BARRIER POLYMERS, TYPES AND PROPERTIES OF HIGH BARRIER POLYMERS, DILUTE SOLUTION ABSORPTION, BARRIER PROPERTIES OF NITRILE POLYMERS AND ACRYLONITRILE-STYRENE COPOLYMERS ARE ALSO CONSIDERED. 12 REFS. (ACS CENTENNIAL MEETING, NEW YORK CITY, APRIL 1976) 9351

ENGINEERING PROPERTIES AND PERFORMANCE OF HIGH ACRYLONITRILE/STYRENE POLYMER SYSTEMS 77-12 05484C

HALL, W. J. CHI, H. K.

CLEVELAND, DHID, DCT.1976, P.1-5. CONFER.012 PUBLCN.
DETAILS- SPE. HIGH PERFORMANCE PLASTICS. NATIONAL TECHNICAL
CONFERENCE

HIGH ACRYLONITRILE-STYRENE COPOLYMER SYSTEMS, CONTAINING 60-70% ACRYLONITRILE, ARE DISCUSSED WITH PARTICULAR REFERENCE TO SYNTHESIS, THERMAL STABILITY, MELT VISCOSITY, GENERAL PROPERTIES OF AN INJECTION MOULDED SPECIMEN, ORIENTED FILM AND SHEET PROPERTIES, MELT FLOW, CHEMICAL RESISTANCE UV RESISTANCE, DUTDOOR WEATHERING RESISTANCE AND PERMEABILITY TRANSFER OF GASES AND WATER VAPOUR. 8 REFS. 42C391C21-9

CHROMATOGRAPHIC CETERMINATION OF THE GAS PERMEABILITY OF FILMS 77-12 05404L

ARKHIPTSEV, N. E. SHISHIMAROV, A. A. MOCHALOVA, L. A.

PLAST.MASSY, NO.4, 1977, P. 73-4

RUSSIAN. THE METHOD PERMITS DETERMINATION OF THE PERMEABILITY OF POLYMERIC FILMS TO CARBON DIOXIDE, OXYGEN AND NITROGEN. 3 REFS. 93513T SELECTED TRANSLATIONS FROM THIS JOURNAL ARE AVAILABLE TO SUBSCRIBERS TO THE RAPRA PRODUCED INTERNATIONAL POLYMER SCIENCE AND TECHNOLOGY

STUDY OF POLYMERS BY INVERSE GAS CHROMATOGRAPHY 77-01 00297L

BRAUN, J. M. GUILLET, J. E.

ADV.PDLYM.SCI.,NO.21,1976,P.107-45

GAS CHROMATOGRAPHY; REVIEW. 116 REFS. 91T

PERMEABILITY -STRUCTURE RELATIONSHIPS OF HIGH POLYMERS 77-01 00337C

SALAME, M.

COATINGS PLAST.PREPRINTS,36,NO.1,AFR.1976,P.488-97.CONFER.PUBLCN. DETAILS- 1ST CHEMICAL CONGRESS ON N.AMERICAN CONTINENT, MEXICO CITY, DEC.1975

GAS PERMEABILITY; CHEMICAL STRUCTURE; LIQUID. 10 REFS. 9351

EFFECTS OF GASEOUS ENVIRONMENTS ON POLYMERS 77-03 01455L BROWN, N.

MAT.SCI.ENGNG.,25,NO.1/2,SEPT./OCT.1976,P.87-91

THIS PAPER EXAMINES THE EFFECT OF GASES SUCH AS NITROGEN AND ARGON. ON CRAZING AND MECHANICAL PROPERTIES OF POLYMERS. 19 REFS. 93513

GAS PERMEATION OF POLYMER BLENDS. III. POLYVINYL CHLORIDE (PVC) / CHLORINATED POLYETHYLENE (CPE) 77-07 02524L

SHUR, Y. J. RANBY, B.

J.APPL.POLYM.SCI.,20,NO.11,NOV.1976,P.3105-19

RESULTS OF MEASUREMENTS OF TRANSPORT OF GASES (HELIUM, DXYGEN, NITROGEN AND CARBON DIDXIDE) THROUGH BLENDS OF PVC AND CPE ARE PRESENTED AND THE DATA RELATED TO THE COMPATIBILITY OF THE BLENDS. THE CIFFUSIVITY AND SOLUBILITY OF THE GASES ARE DISCUSSED ON THE BASIS OF THE FREE VOLUME CONCEPT. 38 REFS. 42C382-6125-93513

GAS PERMEATION OF POLYMER BLENDS. IV. POLYVINYL CHLORIDE (PVC) /ACRYLCNITRILE-BUTADIENE-STYRENE (ABS) TERPOLYMER 77-07 02525L

SHUR, Y. J. RANBY, B.

J. APPL.POLYM.SCI.,20,NO.11,NOV.1976,P.3121-31

THE TRANSPORT BEHAVIOUR OF HELIUM, CXYGEN, NITROGEN AND CARBON DIOXIDE IN MEMBRANES OF PVC/ABS BLENDS WAS STUDIED AT 25C. 19 REFS. (PT.III, IBID, P.3105-19). 42C382-6125-93513

MODEL INCORPORATING REVERSIBLE IMMOBILISATION FOR SORPTION AND DIFFUSION IN GLASSY POLYMERS 7401-20800L

TSHUDY, J.A. VONDFRANKENBERG, C.

J.POLYM.SCI., POLYM.PHYS.ED., 11, NO.10, OCT.1973, P.2027-37

A MODEL INCORPORATING REVERSIBLE, BIMOLECULAR IMMOBILISATION FOR DIFFUSION AND SORPTION IN GLASSY POLYMERS IS DEVELOPED. SORPTION IS CONSIDERED TO OCCUR BY TWO DISTINCT MECHANISMS: ORDINARY DIFFUSION-CONTROLLED SORPTION AND SORPTION RESULTING FROM THE IMMOBILISATION OF DIFFUSING GAS MOLECULES BY PREEXISTING SITES IN THE POLYMER. EXPRESSIONS ARE OBTAINED FOR EQUILIBRIUM SORPTION, TRANSIENT SORPTION, AND TIME LAG. THE EFFECTS OF KINETIC PARAMETERS OF THE MODEL ARE ILLUSTRATED AND DISCUSSED. 15 REFS. 93513

ABSORPTION KINETICS OF SOLVENT VAPOURS IN POLYMER FILMS 7408-25139L

GLOECKNER, G. LCHMANN, D.

PLASTE U.KAUT., 21, NO.5, MAY 1974, P.354-8 GERMAN.

SORPTION OF METHYLENE CHLORIDE AND ACETONE VAPOURS BY FILMS OF PS AND SAN WITH 37.6 AND 60.1 MCL % ACRYLONITRILE, WAS INVESTIGATED USING THE PRAGER AND LONG APPARATUS. THE REDUCED SORPTION CURVES FOR THE FIRST PRESSURE INTERVAL SHOW THE KNOWN CHARACTERISTIC OF ANOMALOUS SORPTION. ALL SORPTIONS OF METHYLENE CHLORIDE VAPOUR BY THE SAN FILMS CONTAINING 37.6% ACRYLONITRILE UP TO 60 MICRONS THICK GAVE THE SAME DIFFUSION COEFFICIENTS IN THE FIRST PRESSURE INTERVAL, BUT THE OTHER FILMS GAVE LOWER VALUES IN ALL PRESSURE INTERVALS. 49 REFS. 42C21-625-93513

EFFECT OF LIQUID PHASE CONTACT ON THE PERMEATION OF GASES THROUGH POLYMERIC FILM/LIQUID LAYER COMPOSITES 7409-25723L

TALWAR. R.

DISS.ABS.INT.,B,34,NO.10,APRIL 1974,P.4947

THE TRANSMISSION OF CARBON DIOXIDE THROUGH POLYETHYLENE/LIQUID (WATER AND SODIUM CHLORIDE) COMPOSITES WAS STUDIED IN ORDER TO DETERMINE HOW THE PRODUCT INFLUENCES GAS OR VAPOUR TRANSMISSION THROUGH PACKAGING MATERIALS. (THESIS, RUTGERS UNIVERSITY, THE STATE UNIVERSITY OF NEW JERSEY, 1973, PP.151, ORDER NO. 74-8824) 42C11-6P1-93513

SIMPLE RADIOACTIVE TRACER METHOD FOR STUDYING SORPTION AND DIFFUSION OF VAPOURS IN POLYMERS 7510-37548L

DYER, A. NEWNS, A.C.

EUR.POLYM.J.,11,NC.5/6,MAY/JUNE 1975,P.397-8

A SIMPLE RADIDACTIVE METHOD FOR THE STUDY OF ISOTOPIC DIFFUSION OF SMALL MOLECULES IN POLYMERS IS DESCRIBED AND THE RESULTS OF AN EXPERIMENT USING A CELLOPHANE SAMPLE ARE GIVEN. 4 REFS. 93513T

MASS-SPECTROMETRIC METHOD FOR INVESTIGATION OF GAS PERMEABILITY OF PCLYMER FILMS 73-05 12536L

IVASHCHENKO, D. A. KROTOV, V. A. TALAKIN, D. G. FUKS, E. V.

VYS.SOED., 14A, NO.9, SEPT. 1972, P. 2109-12

RUSSIAN. THE METHOD, WHICH ALSO ALLOWS DIFFUSION AND SOLUBILITY CONSTANT MEASUREMENTS, WAS TESTED ON SILICONE RUBBER SAMPLES. SOLUBILITY COEFFICIENTS DETERMINED BY TWO OTHER METHODS AGREED WITH THE VALUE FROM THE MASS SPECTROMETRIC METHOD WITHIN THE LIMITS OF EXPERIMENTAL ERROR. 2 REFS. 45C-93513

DIFFUSIVE AND BULK FLOW TRANSPORT IN POLYMERS 73-06 13300L

YASUDA, H. PETERLIN, A.

J.APPL.POLYM.SCI.,17,NO.2,FEB.1973,P.433-42

THE TRANSPORT OF VARIOUS PERMEANTS (GAS, DISSOLVED GAS, LIQUID SOLVENT AND SOLUTE) THROUGH POROUS AND HOMOGENEOUS (NON-POROUS) POLYMER MEMBRANES UNDER THE DRIVING FORCES OF PRESSURE GRADIENT AND CONCENTRATION GRADIENT IS DISCUSSED. IN THE ABSENCE OF A PRESSURE GRADIENT, THE TRANSPORT PROCESS MAY BE DESCRIBED AS DIFFUSION. IN THE PRESENCE OF A PRESSURE GRADIENT, THE TRANSPORT PROCESS MAY BE DIFFUSION AND/OR BULK FLOW, DEPENDING ON MEMBRANE STRUCTURE AND NATURE OF PERMEANT. 16 REFS. 6M-9351

STUDMES OF DIFFUSION IN POLYMERS BY GAS CHROMATOGRAPHY 73-08 14644L

GRAY. D. G. GUILLET, J. E.

MACROMOLECULES, 6, NO. 2, MARCH/APRIL 1973, P. 223-7

IN THIS PAPER THE FACTORS WHICH GOVERN PEAK SHAPE IN GAS CHROMATOGRAPHY ARE OUTLINED AND THEIR RELEVANCE TO DIFFUSION MEASUREMENTS ON PCLYMERIC SUBSTRATES IS INVESTIGATED. BY A SUITABLE CHOICE OF CONDITIONS THE VAN DEEMTER EQUATION ENABLES DIFFUSION COEFFICIENTS TO BE CALCULATED FROM THE VARIATION IN CHROMATOGRAPHIC PEAK WIDTH WITH CARRIER GAS FLOW RATE. THIS METHOD IS APPLICABLE TO SOME HYDROCARBON PENETRANTS IN A PE STATIONARY PHASE. 20 REFS. 93513

POLYMERS WITH HIGH GAS IMPERMEABILITY 73-10 16349L

PELLINO, E.

MAT.PLAST.ELAST.,39,NO.8,AUG.1973,P.623-6

ITALIAN. THE ARTICLE DEALS WITH THE USE OF PLASTICS MATERIALS AS A SUBSTITUTE FOR GLASS IN THE PACKAGING OF

CARBONATED DRINKS. THE BASIC REQUIREMENTS OF THE MATERIAL ARE CONSIDERED: IMPERMEABILITY TO CARBON DIOXIDE, DXYGEN AND WATER, MECHANICAL AND THERMAL PROPERTIES IN GENERAL, AND IMPACT RESISTANCE. CURRENT APPLICATIONS AND POSSIBLE DEVELOPMENTS ARE DISCUSSED, WITH EXAMPLES FROM EUROPE AND THE USA. THE RESPONSE OF THE GLASS INDUSTRY IN THE FACE OF COMPETITION FROM PLASTICS IS ALSO DISCUSSED. 10 REFS. 93513 6P21-93513

DIFFUSION TIME LAG IN HETEROGENEOUS POLYMER MEMBRANES 73-12 17060L

KEMP, D. R.

DISS.ABS.INT.,B,33,NO.7,JAN.1973,P.3056

THE DIFFUSION TIME LAG WAS STUDIED IN POLYMERIC MEMBRANES CONTAINING A DISPERSED SECOND PHASE WHICH IMMOBILISED PART OF THE DIFFUSING SUBSTANCE. EQUATIONS WERE DEVELOPED FOR PREDICTING THIS CUANTITY BY OBTAINING ASYMPTOTIC SOLUTIONS OF THE APPROPRIATE TRANSPORT EQUATIONS. A CLOSED-VOLUME, PRESSURE-GROWTH METHOD WAS USED TO COLLECT TIME LAG AND PERMEABILITY DATA FOR TRANSPORT OF CARBON DIOXIDE, METHANE, NITROGEN AND HELIUM THROUGH SILICONE RUBBER MEMBRANES CONTAINING VARIOUS AMOUNTS OF 5A MCLECULAR SIEVE CRYSTALS. SORPTION CHARACTERISTICS OF THE MEMBRANES FOR THE GASES WERE MEASURED GRAVIMETRICALLY. RESULTS SHOWED THAT THE TIME LAG WAS INCREASED BY UP TO TWO ORDERS OF MAGNITUDE BY DESPERSING RELATIVELY SMALL AMOUNTS OF MOLECULAR SIEVE IN THE POLYMER. (THESIS, UNIVERSITY OF TEXAS AT AUSTIN, 1972, PP.187, ORDER NO.73-466) 45C-6M-93513

SPECTROPHOTOMETRIC DETERMINATION OF AVERAGE CONCENTRATION OF SULPHUR DIOXIDE IN AIR BY PERMEATION THROUGH POLYMER MEMBRANES 73-12 17061L

REISZNER, K. D.

DISS.ABS.INT.,B,33,NO.8,FEB.1973,P.3523

A SIMPLE, LOW-COST METHOD FOR MEASUREMENT OF THE AVERAGE CONCENTRATION OF SULPHUR DIOXIDE IN AIR IS DESCRIBED. IT CONSISTS OF A SIMPLE SAMPLING PROCEDURE COUPLED WITH THE

WEST-GAEKE METHOD FOR SUBSEQUENT DETERMINATION OF THE TRAPPED SULPHUR DIDXIDE. ESSENTIALLY, A GLASS TUBE, STOPPERED AT ONE END. IS FILLED WITH A SOLUTION OF 1.0M SODIUM TETRACHLOROMERCURATE AND THE EXPOSED END COVERED WITH A SILICONE RUBBER MEMBRANE. WHEN THE TUBE IS EXPOSED TO AIR CONTAINING SULPHUR DIOXIDE, THE SULPHUR DIOXIDE PERMEATES THROUGH THE MEMBRANE AND IS ABSORBED BY THE SOLUTION. (THESIS, LOUISIANA STATE UNIVERSITY AND AGRICULTURAL AND MECHANICAL COLLEGE, 1972, PP.89, ORCER NO.73-2980) 45C-6M-93513

REACTION OF NITROGEN DIOXIDE WITH NYLON 66 73-12 17253L

ITOH, Y. JELLINEK, H. H. G. YOKOTA, R.

POLYM.J. (JAP.) .4.NO.6,1973,P.601-6

CHAIN SCISSION OF NYLON 66 FILMS DUE TO EXPOSURE TO NITROGEN DIOXIDE AT A PRESSURE OF 0.5 MM.HG WAS STUDIED AS A FUNCTION OF TEMP., FILM THICKNESS AND POLYMERMORPHOLOGY. THE RATE OF RANDOM CHAIN SCISSION WAS VERY SENSITIVE TO CHANGES IN MORPHOLOGY CAUSED BY VARIATIONS IN SOLVENT COMPOSITION FOR FILM CASTING AND IN METHOD OF PREPARATION. DECREASE OF CHAIN SCISSION WITH FILM THICKNESS INDICATED THAT DIFFUSION OF NITROGEN DIOXICE INTO NYLON 66 WAS RATE-DETERMINING. AMIDE LINKS IN THE CHAIN FOLDS OF THE POLYMER MOLECULES LOCATED IN THE INTERFACIAL REGION BETWEEN AMORPHOUS AND CRYSTALLINE POLYMER PORTIONS WERE THOUGHT TO BE PARTICULARLY SUSCEPTIBLE TO CHAIN SCISSION, DUE TO THEIR STRAIN ENERGY. 5 REFS. 43C313-93513

REACTIONS OF LINEAR POLYMERS WITH NITROGEN DIOXIDE AND 73-12 17421L SULPHUR DIOXIDE

JELLINEK, H. H. G.

TEXT.RES.J.,43,NC.10,OCT.1973,P.557-60

THE EFFECT OF NITROGEN DIOXIDE AND SULPHUR DIOXIDE ON LINEAR POLYMERS IN PRESENCE AND ABSENCE OF AIR, DZUNE AND NEAR-UV IRRADIATION WAS STUDIED. BESIDES RANDOM CHAIN SCISSION AND CROSSLINKING, INCCRPORATION OF RESPECTIVE GROUPS ALONG THE SIDES OF THE CHAINS WAS FOUND TO TAKE PLACE. LOW MOLEC. WT.

COMPOUNDS WERE ALSO EVOLVED. DEGRADATION WAS SENSITIVE TO THE MORPHOLOGY OF THE POLYMER AND ALSO TO THE PERCENTAGE OF CRYSTALLINITY, SIZE AND SIZE DISTRIBUTION OF CRYSTALLITES IN THE POLYMER FILMS. RANDOM CHAIN SCISSION IS DISCUSSED IN DETAIL. 13 REFS. (ACS, DIV. OF CELLULOSE, WOOD AND FIBER CHEMISTRY, SYMPOSIUM ON TEXTILE FINISHING CHEMISTRY, 164TH NATIONAL MEETING, NEW YORK, AUG./SEPT. 1972) 93513

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DETERMINE		25	79-01 27117L
DETERMINE	27117L	26	79-01 27119L
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DETERMINED		30	78-03 12526L
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18 79-12	39758L	34	7401-20800L
DELITEDIUM		35	73-05 12536L
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DEVELOPMENTS	F 22011		73-12 17060L
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DIFFUSION	DIOXIDE
38 73-12 17253L	10 81-01 58208L
	10 81-01 58209L
DIFFUSION COEFFICIENT	37 73-12 17061L
2 82-03 03931L	38 73-12 17253L
4 82-02 02270L	38 73-12 17421L
17 80-06 48299L	
21 79-10 36895L	DISCHARGE
24 79-01 26081L	29 78-05 15701L
32 77-01 00297L	
35 73-05 12536L	DISCUSSED
	19 79-11 38147L
DIFFUSION COEFFICIENTS	
1 82-03 03929L	DOMAINS
8 81-06 67207L	22 79-08 35444L
25 79-01 27117L	
34 7408-25139L	DRAW RATIO
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DIFFUSION CONSTANTS	DSC
7 81-11 76102L	4 82-02 02023L
, 01 11 101021	. 02 02 020232
DIFFUSIVE	DUAL
36 73-06 13300L	7 81-07 69357L
30 13-00 133001	, 010.0,33.2
DIFFUSIVITIES	E.G
2 82-03 03931L	13 80-10 53121A
2 02 03 03/312	23 79-06 32280C
DIFFUSIVITY	23 ,, 00 52200
30 78-03 12526L	EDIN, J. A. D.
33 77-07 02524L	2 82-03 03931L
33 11-01 023242	2 02 03 03/312
DILUENT	EFFECT
3 82-03 03935L	5 82-01 00264L
3 82 03 037336	8 81-06 67207L
DILUT	9 81-05 65408L
23 79-06 32280C	11 80-12 57154L
23 19-00 322000	15 80-07 50126L
DILUTE	17 80-06 479530
DILUTE	17 80-06 47933C
31 78-01 11207L	11 80-00 402336
DINETHIAL CIT DANK CODO! MED	EFFECTS
DIMETHYLSILOXANE COPOLYMER	
6 81-11 76608C	=-
DIOVIDE	33 77 - 03 01455L
DIOXIDE	
1 82-03 03930L	

ELASTIC MODULUS	ENERGY ABSORPTION
19 79-11 38007L	9 81 - 02 59593L
	muo suren Talo
ELASTOMERE	ENGINEERING
13 80-10 53121A	31 77-12 05484C
ELASTOMERIC	ENTHALPY
14 80-09 52281L	1 82-03 03930L
14 60 07 322016	2 82-03 03931L
ELASTOMERS	18 79-12 39757L
13 80-10 53121A	
	ENTROPY
ELECTRICAL	2 82 - 03 03931L
7 81-10 75169L	17 80-04 45453L
1 01 10 131072	28 78-07 18239L
	20 10 01 102572
ELECTRICAL APPLICATION	
12 80-11 55922L	ENVIRONMENTS
	33 77-03 01455L
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9 81-02 59593L	EOUS
9 61-02 373736	14 80-09 526200
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9 81-02 59593L	
19 79-11 38147L	EPDM
26 78-09 21549L	13 80-10 53121A
27 78-08 20094L	CD4
28 78-06 16996L	EPM
	13 80-10 53121A
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20 79-10 36773L	EPOXY GROUP
20 17 10 30.102	3 82-03 03935L
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ELECTRODE	CANATION
12 80-11 55922L	EQUATION
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ELECTRODELESS	2 82 - 03 03931L
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2, 10 0, 23, 020	EQUILIBRIUM
CHROTICONIC	4 82-02 02270L
ENDOTHERMIC	4 02 02 022102
8 81-06 68247L	
	EQUIPMENT
ENERGETICS	2 82-03 03931L
18 79-12 39757L	18 79-12 39758L
	30 78-02 11562L
THEREV	
ENERGY	EDCURY P N
9 81-05 66434L	ERSHOV, B. N.
	6 81-12 79071L

ESTIMATIO	N		FILLERS	
28	78-07	18239L	6	81-11 76608C
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ETC				
14	80-09	52281L	FILM	
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ETHANE			3	82-03 03935L
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7		76102L	17	80-06 47953C
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ETHYLENE	CODOLVE	IED	22	79-08 35301L
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27	18-08	19713L	35	7409-25723L
ETHALENE.	000041 5	NE DUDDED	38	73-12 17253L
		NE RUBBER	20	13.15 115736
13	80-10	53121A	CT) NC	
			FILMS	01.11.7/1031
EVALUATIO			7	81-11 76102L
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	_		9	81-05 65408L
EVAPORATI			11	80-12 57154L
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			15	80-08 51211L
EXOTHERMI	С		16	80-06 48608L
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FABRICS			25	79-01 27117L
22	79-06	32784L	27	78-08 20094L
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FACTORS			32	77-12 05404L
9	81-05	65408L	34	7408-25139L
11		57154L	35	73-05 12536L
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* 2	0,0 01	301202	38	73-12 17421L
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	79-06	227941	FLAME ION	ICATION
22	19-00	321046		82-03 03931L
CICUTAN			4	02-03 03731L
FICKIAN	00 00	020211	CI ANNADTI	TTV
2	82-03	USYSIL	FLAMMABIL	
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FILLERS		000//1		,
5	82-01	0C264L		

FLAMMABILITY	FREE
19 79-11 38007L	9 81 - 05 66434L
., ,, ,, ,,	
FLORY-HUGGINS CONSTANT	FREE ENERGIES
28 78-07 18239L	28 78-07 18239L
20 10 01 102372	
FLOW	FREE RADICALS
23 79-06 32280C	29 78-05 15701L
	2, 10 03 131012
36 73-06 13300L	FREE VOLUME
CLUCTILL TION	3 82-03 03934L
FLUCTUATION OF ((434)	9 81 - 05 66434L
9 81-05 66434L	
FLUORINE	33 77-07 02524L
6 81-12 79071L	
	FREQUENCY
FLUORINE-CONTAINING PCLYMER	24 79-01 26459L
6 81-12 79071L	
	GALVANISATION
FLUX	18 79-12 39758L
10 81-01 58208L	
10 81-01 58209L	GAMMA-IRRADIATION
	7 81-11 76102L
FOAMS	
14 80-09 52620C	GAS CHROMATOGRAPH
	30 78-02 11562L
FOOD PACKAGING	
14 80-09 52281L	GAS CHROMATOGRAPHY
	4 82 - 02 02270L
FOODSTUFFS	7 81-11 76102L
14 80-09 52281L	18 79 - 12 39758L
• ,	24 78-12 24850L
FOOTBALL	28 78-07 18239L
9 81-02 59593L	32 77-01 00297L
, 02 02 373.02	
FORMATION	GAS PERMEABILITIES
4 82-02 02023L	29 78-05 15701L
1 02 02 020232	
FORMULATIONS	GAS PERMEABILITY
9 81-02 59593L	1 82-03 03929L
13 80-10 53121A	1 82-03 03930L
12 OO TO STIFTH	2 82-03 03931L
FRACTURE SURFACES	2 82-03 03932L
	3 82-03 03934L
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GAS PERMEABILITY
GAS PERMEABILITY
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GLOW-DISCHARGE POLYMERISATION
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          78-12 24850L
                                      GRAFT COPOLYMERS
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          77-03 01455L
    33
                                      GRANULES
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GERMAN
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                                                 81-02 59593L
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78-09 21549L
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                                      HEAT CURING
GIVEN
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GLASS TRANSITION TEMPERATURE
                                      HEAT OF SOLUTION
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          79-10 36895L
                                            2
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                                                 82-03 03932L
GLASSY
                                      HEAT TREATMENT
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                                            8 81-06 67207L
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GLC
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                                      HETEROGENEITY
GLOW
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GLOW DISCHARGE POLYMERISATION
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	79-06 32280C	HYDROGEN
	78-06 16996L	17 80-06 47953C
		17 80-06 48299L
	77-12 05484C	21 79-10 36895L
	77-01 00337C	21 79-10 300936
36	73-10 16349L	CYDOCCEN DOND
		HYDROGEN BOND
HIGH-PRES	SSURE	33 77-03 01455L
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		HYDROGEN BONDING
HIGHLY		3 82-03 03934L
3	82-03 03935L	
•	02 03 03/372	HYDROLYSIS
HOLE FORM	AATTON	6 81-11 76608C
	82-03 03932L	0 01 11 .00000
2	82-03 U3932E	10
HOLES		16 80-06 49196L
2	82-03 03932L	
		II
HOMOGENE 1		7 81-07 69357L
19	79-11 38007L	10 81-01 58209L
		11 80-12 57154L
HOMOPOLYM	1ERS	
9	81-05 66434L	III
•		9 81-05 65408L
HOPFENBER	ос. н. в.	33 77-07 02524L
5	82-01 0C520L	
יכ	62-01 0CJZQL	IKEDA, R. M.
	V 14	4 82-02 02023L
HUANG, R.		4 02-02 020231
7	81-11 76102L	T.W.CO.T. T.C.F.D.
		IMMOBILISED
HUMIDITY		1 82-03 03930L
19	79-11 38147L	
		IMPACT MODIFIERS
HUNGARIAN	1	14 80-09 52281L
8	81-06 68247L	**
Ū		IMPACT PROPERTIES
HYDROCARE	RAN	19 79-11 38007L
	82-03 03929L	
_	73-08 14644L	IMPACT RESISTANCE
36	13-00 T4044F	14 80-09 52281L
	2016	36 73-10 16349L
HYDROCARE		30 /3-10 10349L
2	82-03 03931L	

INTERACTION PARAMETERS IMPERMEABILITY 82-02 02270L 73-10 16349L **INTERACTIONS** IMPORTANCE 82-03 03934L 78-08 20094L 27 82-02 02270L IMPURITY INTERFACE 82-02 02270L 79-11 38007L 19 INCLUDE INTERMOLECULAR INTERACTION 29 78-06 16346L 82-02 02270L INCLUDED INTERNAL STRESS 78-06 16996L 28 78-05 15701L 29 78-06 16346L 29 INCLUDING 81-11 76608C 6 INTERPENETRATING POLYMER NETWO 79-11 38007L INCOMPATIBILITY 79-11 38007L INTRINSIC VISCOSITY 82-02 02023L 4 INCREASE 15 80-07 50126L 82-03 03932L INTRODUCTION INDIA 79-01 26881L 25 79-01 26881L 25 INVERSE INFINITE DILUTION 82-02 02270L 4 78-07 18239L 77-01 00297L 32 INFLUENCE 81-12 79071L INVESTIGATED 6 78-06 16346L 29 80-06 49196L 16 79-10 36895L 21 INVESTIGATION 35 73-05 12536L INJECTION MOULDED 77-12 05484C IR SPECTRA 81-11 76102L INJECTION MOULDING 81-10 75169L IR SPECTROSCOPY 3 82-03 03935L INDRGANIC 78-09 20897L 26 IRRADIATED 6 81-12 79071L INTERACTION 26 78-09 20897L

IRRADIATION		KANEGAFUCHI CHEMICAL INDUSTRY
	57154L	7 81-10 75169L
	47953C	
-	36776L	KANEKA PARNEX
21 79-10	367762	7 81-10 75169L
TODARYATION COO	CCL TNIK TNC	7 01 10 131072
IRRADIATION CRO		MARIAT T
7 81-11	76102L	KAWAI, T.
		6 81 - 11 77873L
ISOBUTANE		W = 1.15 = T = 6
2 82-03	03931L	KINETIC
		4 82-02 02270L
ISOMERISM	•	34 7401-20800L
21 79-10	36895L	
		KINETICS
ISOTHERM		9 81 - 05 66434L
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		34 7408-25139L
ISOTHERMS		,
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ISOTOPE	· · · · ·	
17 80-06	492001	KOROS, W. J.
17 80-08	40299L	5 82-01 00520L
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ITALIAN	000071	MONACC D
	38007L	KOVACS, B.
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36 73- 10	16349L	vim 11 m
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		KRYSZTAFKIEWICZ, A.
IX		5 82-01 00264L
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26 78-09		LAMINATE
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KAMIDE, K.		-
6 81-11	778731	LAMINATED
0 01 11	, , , , , , , , , , , , , , , , , , , ,	25 79-01 26881L
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9 81-05	65408L	
		25 79 - 01 27117L

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LAYER	01 01 603001	MAIK, M.	
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		יכ	02-01 00204L
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30	78-02 11562L		
30	78-03 12526L	MALEIC AN	HYDRIDE
		3	82-03 03935L
LEE, W.	M _	-	
	82-03 03934L	MANABE, S	
,	02 03 03/3/2		81-11 77873L
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	GRADATION	MANUFACTU	prpc
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13	80-10 53121A	2	82-03 03931L
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LONCO		MEASUREME	NT
LOWER	70-09 354441	9	81-05 65408L
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MACDONALI		20	79-10 36773L
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۷	02 03 03/322		-12 24850L
MECHANICA	L PROPERTIES		-01 26459L
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36	73-10 163496	METALLISED	
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		•	-01 26881L
MECHANISM			-12 17060L
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MELT ELOU		METHOD	
MELT FLOW			-02 02270L
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MELT VISC			-01 27119L
31	77-12 05484C		-05 12536L
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MEMBRANE	81-11 77873L	METHODS	
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12	80-11 55922L	20 17	TO JOILDE

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	82-02	02270L	MOLECULAR 2		1 03932L
MIGRATION 4	82-02	02270L	MOLECULAR 13		ATION 54673L
-	82-01	00520L	MOLECULAR 37		17060L
MIXING 13	80-10	53121A	MOLECULAR 2	82-03	03931L
MIXTURES 20	79-10	37611L	2 3 3	82-03	03932L 03934L 03935L
MODEL			9		65408L
	82-03		11		57154L 53121A
	82-03		13 17		45453L
5	82-01		23		32280C
10	81-01	58209L	23	19-00	322000
MODELS			MOLECULAR	WEIGHT	
20	79-10	36773L	11	80-12	57154L
			27	78-08	20094L
MODIFIED					
7	81-11		MONOMERS		5 01011
9	81-05	65408L	13	80-10	53121A
MOISTURE	ARSORPT	TON	MORPHOLOG	TCAL PR	OPERTIES
	82-02		23		32280C
•	0 L 0 L				19713L
MOISTURE !	RESISTA	NT			
4	82-02	02023L	MORPHOLOG		
			5		00264L
MOLEC.WT			13	80-10	53121A
15	80-07	50126L			

MORPHOLOGY		NITROGEN		
	38007L	7		76102L
38 73-12	17421L	8		67207L
		16		48962L
MUNDAY, K.		18		39758L
1 82-03	03929L	23		32280C
		27		19713L
MWD		29		16346L
13 80-10	53121A	32		05404L
15 80-07	50126L	33		01455L
		33		02524L
MYLAR		33		02525L
17 80-06	47953C	37		17060L
		38		17253L
N-HEXANE		38	73-12	17421L
2 82-03				
		NITROGEN C		
NAPP, S. J.		29	78-06	16346L
7 81-07				
9 81-05	66434L	NMR		
		3	82-03	03935L
NEON				
21 79-10	36895L	NOBLE	7 10 00	
		30	18-03	12274L
NEOPENTANE	000011	NOUNT T		
2 82-03	03931L	NOHMI, T.	01-11	77873L
NETHON!		6	91-11	11013L
NETWORK	020251	NON-COROLI	•	
3 82-03	03935L	NON-POROUS		1 22001
NITONEL		36	75-00	13300L
NICKEL	207501	NON-TYRE		
18 79-12	39758L	NUNTITEE 9	91-02	59593L
NITRILE		7	01-02	7777JL
14 80-09	E22011	NOVEL		
14 80-09	52201L		79-10	36776L
NITRILE POLYMER		24		26081L
· · · · · · · · · · · · · · · · · · ·	32280C	24	, 01	200016
23 79-06	322600	NR		
NITRILE POLYMERS	•	11	80-12	57566L
	, 11207L	**	JU 12	J. J. J. O. C.
31 10-01	TILVIL	NYLON		
NITRILE RUBBERS		25	79-01	26881L
	27117L	38		17253L
23 19-01	6 1 A E E	30	16	

OBTAINED			OXYGEN
7	81-11	76102L	3 82-03 03934L
•	.	,	3 82 - 03 03935L
OLENINA,	7. K.		4 82-02 02023L
6		79071L	8 81-06 68247L
O	01 12	190112	16 80-06 48962L
OL TODMER			18 79-12 39758L
OLIGOMER	00-06	401041	21 79 - 10 36776L
16	80-08	49196L	23 79 - 06 32280C
OUE COMBO	3 & (T &) T		24 79-01 26081L
ONE-COMPO		7//006	27 78-08 19713L
6	81-11	76608C	32 77-12 05404L
			33 77-03 01455L
ONTO			
3	82-03	03935L	
			33 77-07 02525L
OPPORTUNI			36 73 - 10 16349L
14	80-09	52281L	
			OXYGEN BARRIER
ORGANIC			12 80-11 55870L
20		36773L	
26	78-09	20897L	OXYGEN PERMEABILITY
26	78-09	21549L	3 82-03 03935L
29	78-05	15701L	4 82 - 02 02023L
ORIENTATI	LON		OZONE
12		55870L	19 79-11 38147L
14		52281L	38 73-12 17421L
•			
ORIENTED			OZONE DEGRADATION
14	80-09	52281L	19 79-11 38147L
17		45453L	
31		05484C	OZONE RESISTANCE
J1	•••	0 2 . 0 . 0	9 81-02 59593L
OTHER			•
9	81-02	59593L	P.3139-46
13		53121A	29 78-05 15701L
19		38147L	27 10 05 151012
19	13-11	JOITIL	PACE, R. J.
OUT			2 82 - 03 03932L
OUT		7/1021	2 02 03 03/326
7	81-11	76102L	PACKAGING
OXIDATION		7//005	4 82-02-02270L 7 81-10 75169L
6		76608C	
26	78-09	21549L	22 79-08 35301L

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PEPPAS, N. A.
PACKAGING
                                                 81-05 66434L
          79-06 32280€
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          78-01 11207L
    31
                                      PERFORMANCE
          73-10 16349L
    36
                                           31 77-12 05484C
PACKAGING APPLICATIONS
                                      PERMACHLOR VALUE
          82-03 03934L
     3
                                                79-06 32280C
                                           23
PACKAGING FILM
                                      PERMEABILITY
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                                            3
                                                 82-03 03934L
                                            6
                                                 81-11 76608C
PAN
                                                 81-12 79071L
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          78-05 15701L
                                                 81-10 75169L
                                            7
                                                 81-11 76102L
                                           7
PARAMETERS
                                                 81-06 67207L
          78-07 18239L
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                                                 81-05 65408L
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                                                 80-12 57154L
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PARTICULARLY
                                                 80-12 57566L
                                           11
         80-12 57566L
    11
                                                 80-11 55870L
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                                                 80-10 53121A
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PARTITION COEFFICIENT
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                                                 80-08 51719L
PARYLENE
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PATENTS
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                                                 79-10 36776L
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                                           22
                                                 79-06 32784L
          82-02 02270L
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                                                 79-02 28108L
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                                                 79-01 27117L
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PENETRANTS
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                                                 78-03 12274L
PENETRATION
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                                           32
                                                 77-01 00337C
PEPPAS. N. A.
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PETP
PERMEABILITY
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          73-05 12536L
          73-10 16349L
    36
                                      PETROL CAPS
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                                                81-10 75169L
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PERMEABILITY COEFFICIENT
                                      PETROLEUM
    22
          79-06 32784L
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                                           6
PERMEABLE
                                      PHARMACEUTICAL APPLICATION
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                                                81-02 59593L
PERMEATED
          79-12 39758L
                                      PHASE
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PERMEATION
                                      PHASE SEPARATION
          82-03 03930L
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          82-03 03932L
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          82-01 00520L
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          80-06 48299L
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                                                78-05 15701L
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          77-07 02525L
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                                      PLASMA POLYMERS
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PEROXIDES
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          78-09 21549L
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                                                82-02 02023L
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                                                82-02 02270L
PETP
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          82-03 03930L
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                                                81-07 69357L
          82-03 03932L
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                                                81-11 76102L
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    18
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13	PLASTIC		POLY-P-XYLYLENES
17		80-11 54673L	27 78-08 20094L
POLYACETYLENE			
26 79-01 27119L 27 78-08 19713L 28 78-07 18239L 30 78-03 12274L 30 78-03 12526L 32 77-01 00297L 33 77-03 01455L PLASTICISER/PVC 12 80-11 55922L PLASTICS 9 81-02 59593L 25 79-01 26881L 26 79-01 26881L 27 79-01 26881L 28 79-01 26081L POLYCARBONATE 1 82-03 03931L 25 79-01 26881L 26 79-01 26081L 27 78-08 19713L POLYCARBONATE 1 82-03 03931L 25 79-01 26881L 26 79-01 27117L POLAR 24 79-01 26081L POLYCARBONATE 1 82-03 03931L 25 79-01 26081L 26 79-01 27117L POLAR 27 78-08 19713L POLYCARBONATE 2 82-03 03931L 3 82-03 03934L POLYCHLOROPRENE 25 79-01 27117L POLAR 3 82-03 03934L POLYCONDENSATION 4 82-02 02023L 7 8-08 19713L POLYCISPERSITY 15 80-07 50126L POLYCIPERSITY 16 8012L POLYCIPERSITY 17 80-08 1	_		POLYACETYLENE
27			
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30			POLYACRYLATE
30			
## POLYAMIDE ## 82-02 02023L ## 82-03 03929L ## 82-03 03929L ## 82-03 03929L ## 82-03 03929L ## 82-03 03933L ## 82-03 03933L ## 82-03 03934L ## 82-01 00520L ## 82-02 02023L ## 82-03 03934L ## 82-02 02023L ## 82-03 03934L ## 82-02 02023L ## 82-03 03934L ## 82-03 03934L			.0 0.02 20200
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11 80-12 57154L POLYESTER-URETHANES 15 80-07 50126L 22 79-08 35444L POLMANTEER, K. E. 6 81-11 76608C 20 79-10 37611L POLY 4 82-02 02023L 27 78-08 19713L POLYETHER-URETHANES 22 79-08 35444L POLYETHER-URETHANES 22 79-08 35444L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L			
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POLMANTEER, K. E. 6 81-11 76608C POLYETHER 20 79-10 37611L POLYETHER-URETHANES 22 79-08 35444L 27 78-08 19713L POLYETHER-URETHANES 22 79-08 35444L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L			22 79-08 35444L
6 81-11 76608C 20 79-10 37611L PULY POLYETHER-URETHANES 4 82-02 02023L 22 79-08 35444L 27 78-08 19713L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L		00 0. 301202	
6 81-11 76608C 20 79-10 37611L PULY 4 82-02 02023L 22 79-08 35444L 27 78-08 19713L POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L	POLMANTER	ER. K. E.	POLYETHER
POLY POLYETHER-URETHANES 4 82-02 02023L 22 79-08 35444L 27 78-08 19713L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L			20 79-10 37611L
4 82-02 02023L 22 79-08 35444L 27 78-08 19713L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L	•		
4 82-02 02023L 22 79-08 35444L 27 78-08 19713L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L	POLY		POLYETHER-URETHANES
27 78-08 19713L POLYETHYLENE POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L		82-02 02023L	22 79-08 35444L
POLYETHYLENE POLYETHYLENE 7 81-11 76102L	•		
POLY-P-PHENYLENETEREPHTHALAMID 7 81-11 76102L	•••		POLYETHYLENE
	POLY-P-PE	HENYLENETEREPHTHAL AMID	7 81-11 76102L

POLYETHYL	ENE	POLYMER	
. 28	78-06 16996L	21	79 - 10 36895L
30	78-03 12526L	22	79-06 32784L
33	77-07 02524L	24	79-01 26081L
35	7409-25723L	27	78-08 19713L
		28	78-07 18239L
POLYGLYCI	DYL ACRYLATE	30	78-02 11562L
3	82-03 03935L	30	78-03 12274L
_		31	77-12 05484C
POLYGLYCI	DYL METHACRYLATE		77-07 02524L
3	82-03 03935L	33	77-07 02525L
•		34	7408-25139L
POLYIMIDE		35	73-05 12536L
8	81-06 67207L		73-12 17060L
· ·	01 00 0.20.2	37	73-12 17061L
POLYISOBU	TYLENE	38	73-12 17421L
11	80-12 57566L		
**	00 12 3.3002	POLYMER-M	IGRANT
POLYISOPR	ENE	4	82-02 02270L
25	79-01 27117L		
	,, 02 2.12.12	POLYMERIC	
POLYMER		6	81-11 77873L
2	82-03 03931L	6	81-12 79071L
2	82-03 03932L		
	82-03 03934L	15	
3 5 7	82-01 00520L	15	
7	81-10 75169L		79-08 35301L
7	81-11 76102L		78-12 24850L
8	81-06 67207L		79-01 26459L
9	81-05 65408L		78-09 20897L
10	81-01 58208L	32	77-12 05404L
10	81-01 58209L	35	7409-25723L
11	80-12 57154L		
12	80-11 55922L	POLYMERIC	FILLER
15	80-08 51211L	16	80-06 49196L
16	80-06 48608L		
16	80-06 48962L	POLYMERIC	IMPACT MODIFIER
17	80-06 47953C	9	81-05 65408L
17	80-06 48299L	14	80-09 52281L
18	79-12 39758L		
19	79-11 38007L	POLYMERIC	MODIFIER
20	79-10 36773L	9	81-05 65408L
20	79-10 37611L		
21	79-10 36776L	POLYMERIC	PROPERTY MODIFIERS
		9	81-02 59593L

POLYMER IS	SATION	POLYMERS
4	82-02 02023L	19 79-11 38147L
6	81-11 76608C	21 79-10 36776L
12	80-11 55870L	22 79-08 35444L
13	80-10 53121A	23 79-02 28108L
28	78-06 16996L	23 79-06 32280C
29	78-05 15701L	25 79-01 26881L
		25 79-01 27117L
POLYMERTS	SATION KINETIC	26 78-09 21549L
29	78-06 16346L	26 79-01 27119L
2,	10 00 103 102	27 78-08 20094L
POLYMERTS	SATION KINETICS	30 78-03 12526L
13	80-10 53121A	31 78-01 11207L
		32 77-01 00297L
DOL VMEDTO	SATION MECHANISM	32 77-01 00337C
29	78-05 15701L	33 77-03 01455L
	78-06 16346L	34 7401-20800L
27	10 00 105+0E	35 7510-37548L
DOL VMEDT	SATION TEMP	36 73-06 13300L
4	82-02 02023L	36 73-08 14644L
7	02 02 020232	36 73-10 16349L
POLYMERIS	ten	38 73-12 17421L
8		J0 13 12 11 12 12
8	81-08 662411	POLYMETHACRYLATE
POLYMERMO	IRPHULUCA	17 80-04 45453L
38	73-12 17253L	21 00 01 13 13 02
30	13 12 112332	POLYMETHYL ACRYLATE
POLYMERS		2 82-03 03932L
1	82-03 03929L	21 79-10 36895L
i	82-03 03930L	28 78-07 18239L
	82-03 03931L	
2 2 3	82-03 03932L	POLYOLEFIN
3	82-03 03935L	4 82-02 02270L
5	82-01 00264L	
Ś	82-01 00520L	POLYPHENYL ETHER
5 7	81-07 69357L	20 79-10 37611L
8	81-06 67207L	
9		POLYPHENYLENE ETHER
1ĺ	80-12 57566L	20 79-10 37611L
13	80-10 53121A	
13	80-11 54673L	POLYPHENYLENE TEREPHTHALAMIDE
17	80-04 45453L	4 82-02 02023L
17	80-06 48299L	12 80-11 55870L
18	79-12 39757L	
		POLYPROPYLENE
		6 81-12 79071L

POLYSAR LTD 9 81-02 59593L	POWCER 1 82-03 03929L 8 81-06 68247L
POLYSILOXANE	
24 78-12 24850L	PP
24 10-12 240301	3 82 - 03 03935L
DOL VETVOCHE	28 78 - 06 16996L
POLYSTYRENE	20 /0 00 10 // 00
9 81-05 65408L	DOTA
11 80-12 57154L	PPTA 00 11 FE070L
15 80-07 50126L	12 80-11 55870L
POLYSULPHIDE	PREPARATION
26 78-09 21549L	27 78-08 20094L
POLYURETHANE	PREPARED
22 79-08 35444L	9 81-05 65408L
POLYVINYL	PRESSURE
16 80-06 49196L	1 82-03 03930L
33 77-07 02524L	5 82-01 00520L
33 77÷07 02525L	10 81-01 58208L
55 11 01 025252	10 81-01 58209L
POLYVINYL ACETATE	18 79-12 39758L
	23 79 - 02 28108L
1 82-03 03929L	30 78-02 11562L
TO MUTHIN TOTAL THE THIN OT LONG	
POLYVINYLTRIMETHYLSILOXANE	36 73 - 06 13300L
24 78-12 24850L	DD F 6611D F 6
	PRESSURES
PORE SIZE	1 82-03 03929L
24 79-01 26459L	2 82 - 03 03932L
26 78-09 20897L	20 79 - 10 37611L
	24 78-12 24850L
PORE SIZE DISTRIBUTION	
6 81-11 77873L	PROBLEM
	25 79-01 27117L
POROUS	
6 81-11 77873L	PROCESSIBILITY
24 79-01 26459L	14 80-09 52281L
26 78-09 20897L	
36 73-06 13300L	PROCESSING
J0 15 00 15500E	13 80-10 53121A
POSSIBLE	14 80-09 52281L
	28 78-06 16996L
13 80-11 54673L	20 /0-00 10 9901
00758710457016	DD CID ANE
POTENTIOMETRIC	PRUPANE 1 82-03 03929L
12 80-11 55922L	1 82-03 03929L

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PVAC
PROPANE
                                                 79-10 36895L
                                          21
          81-11 76102L
     7
                                      PVC
PROPERTIES
                                                 82-03 03930L
        82-01 00264L
     5
                                                 81-06 68247L
                                           8
          81-12 79071L
                                           9
                                                 81-05 65408L
          81-02 59593L
                                                 79-01 26881L
          81-05 65408L
                                           25
     9
                                                 77-07 02524L
                                          33
    19
          79-11 38007L
                                           33
                                                 77-07 02525L
          79-11 38147L
    19
    23
          79-06 32280C
                                      PVCBUTADIENE COPOLYMERS
          78-09 21549L
    26
                                              78-08 19713L
    27
          78-08 20094L
                                          27
    28
          78-06 16996L
                                      RATE OF DIFFUSION
          78-05 15701L
    29
          78-06 16346L
                                                82-03 03931L
                                           2
    29
                                           3
                                                 82-03 03934L
          77-12 05484C
PROPERTY MODIFIERS
                                      RAVVE, A.
                                                 82-03 03935L
         79-11 38007L
PROSTHESIS
                                      REACTION
                                                 73-12 17253L
         81-11 76608C
                                          38
                                      REACTIONS
PROTECTIVE COATINGS
                                          38
                                                 73-12 17421L
          80-11 55870L
    12
                                      RECYCLING
PS
                                                79-11 38007L
                                          19
     1
          82-03 03929L
          82-03 03930L
     1
          80-09 52620C
                                      REFS
    14
                                                 82-02 02023L
          79-01 26081L
    24
                                                 82-01 00520L
                                           5
          78-07 18239L
    28
          7408-25139L
                                                 81-06 67207L
    34
                                          11
                                                 80-12 57566L
                                                 80-11 55922L
PT.VIII.IBID
                                          12
         78-05 15701L
                                          13
                                                 80-10 53121A
    29
                                          29
                                                 78-05 15701L
                                          29
                                                 78-06 16346L
PUMPS
          81-10 75169L
                                    REINFORCED
PURITY
                                          22
                                                 79-06 32784L
          78-03 20094L
    27
                                      REINFORCED PLASTIC
                                                79-06 32784L
                                          22
PVAC
     2
          82-03 03932L
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REINFORCED RUBBER 22 79-06 3		RUBBERY 17	80-06	48299L
RELATIONSHIPS 32 77-01 0	0337C	RUPTURE 19	79-11	380 07L
RELATIVE 1 82-03 0	3929L	RUSSIAN 6 15	81 - 12 80 - 08	
RELAXATION 8 81-06 6	7207L	16 16 21	80-06 80-06 79-10	48608L 49196L
RESIN 14 80-09 5	2281L	22 23 24	79-08 79-02 78-12	28108L 24850L
RESULTS 20 79-10 3	6773L	26 32	79 - 01 77 - 12	
RETENTION VOLUME 4 82-02 0	2270L	SAMUELS, 1	M. R. 82-02	02023L
REVIEW 6 81-11 7 15 80-08 5 32 77-01 0	1211L	SAN 9 23 31 34	81-05 79-06 77-12 7408-2	32280C 05484C
RHEOLOGICAL PROPE 6 81-11 7 13 80-10 5	76608C	SANCHEZ,		
RIGID 25 79-01 2	6881L	SEALANTS 9	81-02	59593L
ROOM TEMPERATURE 6 81-11 7		SEALING 13	80-10	53121A
RUBBER 6 81-11 7 9 81-02 5		SEGMENTAL 2		
13 80-10 5 19 79-11 3 22 79-06 3	3121A 88147L	SEGMENTAL 17		48299L
32 77-01 0 RUBBERS		SELECTION 3	82-03	03934L
11 80-12 5	7566L			

SELECTIVE	SILICONE RUBBER
10 81-01 58208L	11 80-12 57566L
10 81-01 58209L	35 73-05 12536L
	37 73-12 17060L
SELECTIVITY	37 73-12 17061L
13 80-11 54673L	
13 00-11 340/36	SILICONE RUBBERS
	6 81-11 76608C
SEM	6 61-11 700050
12 80-11 55870L	ATLEVINE ADDOLVMEDE
	SILCXANE COPOLYMERS
SENICH, G. A.	6 81-11 76608C
4 82-02 02270L	
	SIMPLE
SENSITISER	2 82-03 03932L
21 79-10 36776L	
21 19-10 307701	SIMPLIFIED
OFLICER	20 79-10 36773L
SENSOR	20 19-10 301136
12 80-11 55922L	<u>_</u>
	SINGLE
SHEET	10 81-01 58208L
15 80-08 51211L	
25 79-01 26881L	SINGLET DXYGEN
31 77-12 05484C	21 79-10 36776L
SHEETS	SIZE
15 80-08 51211L	2 82-03 03931L
15 00-00 312111	
CHELE LITER	SMITH, T. L.
SHELF LIFE	8 81-06 67207L
4 82-02 02270L	6 81-00 672012
	CONTUN CHI COIDE
SIDE GROUPS	SODIUM CHLORIDE
2 82 - 03 03932L	35 7409-25723L
SILANE POLYMER	SOLTAN
16 80-06 49196L	14 80-09 52281L
SILANES	SOLUBILITY
16 80-06 49196L	2 82-03 03931L
10 00 00 171702	4 82-02 02270L
CT1 TC1	6 81-11 766080
SILICA	
5 82-01 00264L	
SILICONE	26 79-01 27119L
6 81-11 76608C	28 78 - 07 18239L

SOLUBILITY	SORPTION
33 77-07 02524L	34 7401-20800L
35 73-05 12536L	34 7408-25139L
	35 7510-37548L
SOLUBILITY COEFFICIENTS	37 73-12 17060L
16 80-06 48962L	
21 79-10 36895L	SPANISH
28 78-07 18239L	9 81-02 59593L
35 73-05 12536L	19 79-11 38147L
	_
SOLUTE	SPECIFIC VOLUME
36 73-06 13300L	3 82-03 03934L
SOLUTES	SPECTROPHOTOCHEMICAL
30 78-03 12526L	21 79 - 10 36776L
SOLUTION	SPECTROPHOTOMETRIC
23 79-06 32280C	37 73-12 17061L
31 78-01 11207L	
31 10 01 111	SPECTROPHOTOMETRY
SOLUTION PROPERTIES	21 79-10 36776L
20 79-10 37611L	
32 77-01 00297L	SPORTS SURFACES
32 11 01 00E7TE	13 80-10 53121A
SOLVAY & CIE	
14 80-09 52281L	STANNETT, V.
14 00-07 322012	5 82-01 00520L
SOLVENT	5 02 01 00 200
4 82-02 02270L	STATISTICAL
20 79-10 37611L	2 82-03 03932L
23 79-06 32280C	
25 79-01 27117L	STEADY-STATE
34 7408-25139L	5 82-01 00520L
	02 01 000200
	STIFFNESS
38 73-12 17253L	14 80-09 52281L
COPPTION	14 80-09 322012
SORPTION	STORAGE
1 82-03 03929L	25 79 - 01 26881L
1 82-03 03930L	25 19-01 20001
2 82-03 03932L	STRENGTH
5 82-01 00520L	
7 81-07 69357L	26 78-09 21549L
9 81-05 66434L	CTOPEC ODTENTATION
18 79-12 39757L	STRESS ORIENTATION
	14 80 - 09 52281L

STRESS-S	TRAIN PROPERTIES	SULPHUR
8	81-06 67207L	38 73-12 17421L
	77-03 01455L	
	.,	SURFACE ENERGY
STRUCTURA	Δ1 .	6 81-11 76608C
21	79-10 36895L	29 78-06 16346L
2.1	17 10 300732	27 .000 2007.02
STRUCTURE	=	SURFACE PROPERTIES
	82-03 03931L	32 77-01 00297L
2		33 77-03 01455L
	82-03 03934L	33 11-03 014776
32	77-01 00337C	CHREACE CIRLCIANE
		SURFACE STRUCTURE
STUDIED		8 81 - 06 68247L
8	81-06 67207L	
		SUSPENSION
STUDIES		8 81-06 68247L
27	78-08 19713L	
		SUSPENSION POLYMERISATION
STUDY		8 81-06 68247L
22	79-06 32784L	
25	79-01 27117L	SYNTHESIS
30	78-02 11562L	6 81-11 76608C
32	77-01 00297L	9 81-02 59593L
32	77 01 002712	27 78-08 20094L
STUDYING		29 78-05 15701L
	82-02 02270L	29 78-06 16346L
4		31 77-12 05484C
23	79-02 28108L	31 11-12 03-10-10
		CANTHECTCED
STUD"ES		SYNTHESISED
36	73-08 14644L	4 82 - 02 02023L
STYRENE	COPOLYMER	SYNTHETIC RUBBER
14	80-09 52281L	25 79-01 26881L
SUBSTRATI	Ε	SYSTEMS
4	82-02 02023L	31 77-12 05484C
SUCCINIC	ANHYDRIDE	T HIROTSU
3	82-03 03935L	29 78 - 06 16346L
•		
SULPHUR		TANK LININGS
10	81-01 58208L	9 81 - 02 59593L
10	81-01 58209L	, 02 02 7777
37	73-12 17061L	TECHNIQUE
<i>31</i>	13-12 110016	24 79-01 26081L
		24 17-01 20001L

TECHNOLOGY		TERYLENE	
	76608C	22 79	-06 32784L
TEMP		TEST	
	03931L		-02 02270L
	03934L		-08 51211L
	76102L		-10 37611L
	52620C		-12 24850L
	35444L		-01 27119L
	32280C		-02 11562L
			-12 05404L
	18239L	32 11	IL OUTOTE
	11562L	TOOTING	
-	11207L	TESTING	-0/ (00/31
38 73-12	17253L		-06 68247L
			-06 32784L
TEMPERATURE			-08 35301L
	03931L	23 79	-02 28108L
2 82-03	03932L		•
14 80-09	52281L	TESTS	
17 80-06	48299L	25 79	-01 26881L
TEMPERATURE DEP	ENDENCE	TG	
1 82-03	03929L	-	-03 03929L
	03931L		-03 03931L
	03932L	2 82	-03 03932L
	,		-01 00520L
TEMPERATURES			-10 53121A
	03929L		-06 48299L
	32784L		-08 35444L
	28108L		-06 32280C
	24850L		
	26081L	TGA	
-	200012		-02 02023L
TENSILE			
8 81-06	67207L	THEORIES 9 81	-05 66434L
TENSILE PROPERT	IFS		
· - · · · · · · · · · · · · · · · · · · ·	67207L	THEORY	
8 01 00	012012	7 81	-07 69357L
TENSILE STRENGT	Н	17 80	-04 45453L
14 80-09	52281L		
	01455L	THERMAL	
		8 81	-06 68247L
TERPOLYMER	0.05.05.1	THEOMAL COOK	EDTIES
33 77-07	U2525L	THERMAL PROP	
		19 79	-11 38007L

THERMAL P	PROPERTIES	TOPOLOGY
28	78-06 16996L	3 82-03 03935L
36	73-10 16349L	
		TOXICITY
THERMAL S	STAR II ITY	26 78-09 21549L
	81-02 59593L	
	78-09 21549L	TRACER
31		35 7510-37548L
51	17 12 034040	
THERMODYN	IAMT C	TRACE MARKS
	82-03 03931L	28 78-06 16996L
2	82-02 02270L	20 10 00 207702
	81-05 66434L	TRANSFER
	79-11 38007L	22 79 - 06 32784L
		30 78-02 11562L
28	78-07 18239L	30 10 02 113022
************	CTIC ELACTOMECE	TRANSITION
	STIC ELASTOMERS	1 82-03 03929L
13	80-10 53121A	32 77 - 01 00297L
T1/ - 61/41/7 6 6		32 11-01 00291E
THICKNESS		TRANSITIONS
4	82-02 02023L	2 82-03 03931L
23		2 62-03 037316
27		TO ANCHICCION
31	78-01 11207L	TRANSMISSION 3 82-03 03935L
38	73 - 12 17253L	3 82 - 03 03935L
	·	TRANSMISSION ELECTRON MICROSCO
THIN		
15	80-08 51211L	12 80-11 55870L
		TO ANCO ADAMON
THIN FILM		TRANSPARANCY
10	81-01 58208L	27 78 - 08 20094L
		TO AMEDICAT
THIOKOL		TRANSPORT 01 04 (7227)
26	78-09 21549L	8 81-06 67207L
		10 81-01 58208L
THROUGH		10 81-01 58209L
35	7409-25723L	36 73 - 06 13300L
TIME		TRANSPORT APPLICATION
37	73-12 17060L	25 79 - 01 26881L
TOI, K.		TRIACETATE
1	82-03 03930L	10 81 - 01 58208L
	•	·
TOLUENE		TRIACETATE/POLYACRYLATE
20	79-10 37611L	10 81 - 01 58209L

TRIMETHYL	STI ANE		VAPOUR
16		49196L	35 7409-25723L
10		1,2,00	
TUBES			VAPOUR PERMEABILITY
	78-02	11562L	1 82-03 03929L
30	10 02		2 82-03 03931L
TWO-COMPO	NENT		
6		76608C	VAPOUR-PHASE
· ·	0	, 00000	4 82-02 02023L
UNDER			
	80-12	57154L	VAPCUR-PHASE POLYMERISATION
* *	00 12	J 12 J 1 L	12 80-11 55870L
URETHANE			
5	82-01	00264L	VAPOUR-PRESSURE OSMOMETRY
,	02 01	002012	20 79-10 37611L
υv			
	80-12	57154L	VAPOURS
11	00 - 12	J11346	1 82-03 03929L
UV DEGRAD	MITTAR		5 82-01 00520L
		57154L	15 80-08 51211L
		05484C	20 79-10 36773L
. 31	11 12	034046	34 7408-25139L
UV IRRADI	TATTON		34 7408-25139L
		17421L	35 7510-37548L
56	13 12	114717	
UV STABIL	TTV		VARIOUS
		65408L	9 81-05 65408L
7	01-05	034006	11 80-12 57154L
VACUUM			15 80-07 50126L
7 7	81-11	76102L	23 79-02 28108L
		28108L	
23	17 02	201002	VEHICLE SUSPENSIONS
VACUUM FO	IDMEN		9 81-02 59593L
7		75169L	, 02 02 00000
ŧ	01-10	751076	VINYL
WAN DED L	A PIAAL	TTRACTION	27 78 - 08 19713L
33		01455L	27 70 00 177131
33	11-05	014336	VINYL ACETATE COPOLYMER
VADOLID			27 78-08 19713L
VAPOUR	00-11	55870L	21 10 00 177152
12		38007L	VINYLALKYLDIMETHYL
19 20		37611L	16 80-06 49196L
21		36895L	10 00 00 171701
		27117L	VISCOELASTIC PROPERTIES
25	17-01	21111L	8 81-06 67207L
			0 01-00 015015

VISCOSITY	1	WILLIAMS, M. J. L.
26	78-09 21549L	1 82 - 03 03929L
VISTRON C	ORP	WIRTSCHAFTSVERBAND DER DEUTSCH
14	80-09 52281L	13 80-10 53121A
VOLUME	•	XENON
9	81-05 66434L	30 78 - 03 12274L
VULCANISA	ATION	ZIEGLER, B.
6	81-11 76608C	8 81-06 68247L
13	80-10 53121A	
		42C11
WALLS		28 78-06 16996L
	78-02 11562L	
		42C11-61122-6125-93513
WATER		33 77-07 02524L
	79-06 32280C	
	78-05 15701L	42C11-8953-93513
	78-06 16346L	7 81-11 76102L
	7409-25723L	
36	73-10 16349L	42C11-935T
		4 82 - 02 02270L
WATER PER	RMEABILITY	
	79-06 32280C	4201103311-6125-93513
	78-09 21549L	27 78 - 08 19713L
20		
WATER VAR	POUR	42C12-6M-93513
	82-01 00520L	6 81-12 79071L
	77-12 05484C	
		42C131D12
WATER VAR	POUR PERMEABILITY	9 81 - 02 59593L
	80-08 51211L	
		42C21C391D11-6125-93513
WATER-SOL	LUBLE	33 77-07 02525L
	79-10 36776L	
		42C35-55CAH-93513
WEATHERIN	NG	3 82-03 03935L
	79-11 38147L	
- -		42C351A-9351
WEATHERIN	NG RESISTANCE	7 81-10 75169L
_	77-12 05484C	
		42C35111-9351
WHICH		28 78 - 07 18239L
21	79-10 36776L	

42C382-622-932	43C6-51SS-9				
8 81-06 68247L	5 82-01 00264L				
420391-7223	45 C				
29 78-05 15701L	6 81-11 76608C				
420391-9	45C-6M-93513				
29 78-05 15701L	37 73-12 17060L				
	37 73-12 17061L				
42D1-625-93513					
25 79 - 01 27117L	6A3-93513				
	30 78-03 12274L				
42F1-7223 29 78-05 15701L	6E1-93				
29 78 - 05 15701L	19 79-11 38147L				
42F1-9	2, , , , 12 002				
29 78-05 15701L	6H2-93513T				
	30 78-02 11562L				
42F1A-7223					
29 78-06 16346L	6M-93513				
	5 82-01 00520L 6 81-11 77873L				
43C1-6125-93513	6 81-11 77873L 10 81-01 58208L				
27 78-08 19713L	10 81-01 58208L				
43C112-625-936	10 01 01 302072				
17 80-06 47953C	6P21-93513				
	36 73-10 16349L				
43C12-93513					
2 82-03 03931L	6122 - 9T				
8 81-06 67207L	20 79-10 37611L				
(20242 02512	625-8(11)3464-93513T				
43C313-93513 38 73-12 17253L	18 79-12 39758L				
36 13-12 112536	18 17 12 371302				
43C318-6A31-723					
4 82-02 02023L					
43C4-93513					
8 81-06 67207L					
42052-0					
43C52-9 26 78-09 21549L					
20 10 07 217776					
43C52-91					
26 78-09 21549L					